



ShreeyashPratishthan's

SHREEYASH INSTITUTE OF PHARMACEUTICAL EDUCATION AND RESEARCH

Approved by AICTE, New Delhi, Government of Maharashtra, DTE Mumbai (**DTE Code- 2572**) and
Affiliated to Dr. BabasahebAmbedkar Technological University, Lonere, & Maharashtra State Board of
Technical Education (MSBTE), (**MSBTE Code- 1838**) Mumbai

1.1.1

The Institution ensures effective curriculum planning and delivery through a well-planned and documented process including Academic calendar and Conduct of continuous internal Assessment

INDEX (2021-22)

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डॉ. बाबासाहेब आंबेडकर तंत्रशास्त्र विद्यापीठ, लोणेरे
Dr. Babasaheb Ambedkar Technological University, Lonere

विद्याविहार, लोणेरे-रायगड ४०२ १०३ (महाराष्ट्र) Vidyavihar, Lonere - Raigad 402 103 (Maharashtra)

Tel: (02140) 275142 Student Helpline: 02140-275212

Website: www.dbatu.ac.in, E-mail: registrar@dbatu.ac.in



DBATU/Reg/AC/Pahrma/2022/119

Dated: 06/06 /2022

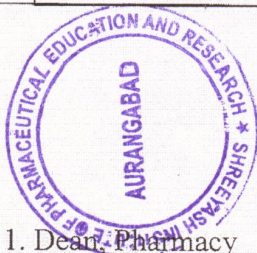
Academic Calendar 2021-22 (EVEN SEMESTER)

B. Pharm First year (Sem - II)

Sr. No.	Activity	Date
1	Commencement of Classes	May 9, 2022
2	First Sessional Examination (Theory)	June 15-21, 2022
3	Exam form filling for Regular & Supplementary Examinations (without late fee)	July 6-11, 2022
4	Exam form filling for Regular & Supplementary Examinations (with late fee)	July 13-17, 2022
5	End of Classes	Aug. 6, 2022
6	Second Sessional Examination (Theory)	Aug. 8-13, 2022
7	Uploading Internal marks to University portal	Aug. 9-16, 2022
8	End Semester Theory Supplementary & Regular Examination	Aug. 17-29, 2022
9	End Semester Practical Examination	Aug. 30- Sept. 5, 2022
10	Commencement of next Semester 2022-23	Sept. 10, 2022

M. Pharm First year (Sem. - II)

Sr. No.	Activity	Date
1	Commencement of Classes	May 11, 2022
2	First Sessional Examination (Theory)	June 15-21, 2022
3	Exam form filling for Regular & Supplementary Examinations (without late fee)	July 6-11, 2022
4	Exam form filling for Regular & Supplementary Examinations (with late fee)	July 13-17, 2022
5	End of Classes	Aug. 6, 2022
6	Second Sessional Examination (Theory)	Aug. 8-13, 2022
7	Uploading Internal marks to University portal	Aug. 9-16, 2022
8	End Semester Theory Supplementary & Regular Examination	Aug. 17-29, 2022
9	End Semester Practical Examination	Aug. 30- Sept. 5, 2022
10	Commencement of next Semester 2022-23	Sept. 10, 2022



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2. University Website



PRINCIPAL
Shreyash Institute Of Pharmaceutical
Education And Research, Aurangabad

(Dr. B. F. Jogi)

Registrar
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Dr. Babasaheb Ambedkar Technological University
LONERE 402 103,
Tal. Mangaon, Dist. Raigad, (Maharashtra)



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Date: 06/06/2022

Academic calendar 2021-22 (EVEN SEMESTER)

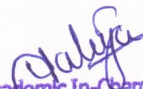
B. Pharm. First Year (Sem-II)

Sr. No.	Activity	Date(s)
1	Commencement of classes	May. 9, 2022
	Cultural Event	May. 18, 2022
2	First Sessional Examination	June. 15-21, 2022
3	Theory exam form filling for regular and supplementary examinations (without late fee)	July. 6-11, 2022
4	Exam form filling for regular and supplementary examinations (with late fee)	June. 13-17, 2022
5	End of Classes	Aug. 6, 2022
6	Second Sessional Examination (Theory)	Aug. 8-13, 2022
7	Uploading internal marks to University portal	Aug. 9-16, 2022
8	End semester theory supplementary and regular examination	Aug. 17-29, 2022
9	End Semester Practical Examination	Aug. 30-Sept. 5, 2022
10	Teachers Day	Sep. 5, 2022
11	Commencement of next semester 2022-23	Sept. 10, 2022

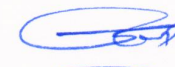
➤ Practical sessional examination should be schedule at Institute level.

List of holidays/ festivals

Sr. No.	Holidays/ festivals	Date
1	Bodh Purina	May. 16, 2022
2	Moharram	Aug. 9, 2022
3	Independence Day	Aug. 15, 2022
4	Ganesh Chaturthi	Aug. 31, 2022


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Tel: (02140) 275142 Student Helpline: 02140 - 275212

Website: www.dbatu.ac.in, E-mail: registrar@dbatu.ac.in

Dr. Bhagawan F. Jogi

Registrar

डॉ. भगवान फ. जोगी

कुलसचिव

क्र. डॉबाआंतवि/कुलसचिव कार्यालय/२०२२/८८

दिनांक : १९/०४/२०२२

Academic Calendar 2021-22 (EVEN SEMESTER)

B. Pharm Second year (Sem. - IV)

Sr. No.	Activity	Date(s)
1	Commencement of Classes	April 18, 2022
2	First Sessional Examination (Theory)	June 6-11, 2022
3	Exam form filling for Regular & Supplementary Examinations (without late fee)	July 6-11, 2022
4	Exam form filling for Regular & Supplementary Examinations (with late fee)	July 13-17, 2022
5	End of Classes	July 27, 2022
6	Second Sessional Examination (Theory)	July 28- Aug. 8, 2022
7	Uploading Internal marks to University portal	Aug. 9-13, 2022
8	End Semester Theory Supplementary & Regular Examination	Aug. 6-20, 2022
9	End Semester Practical Examination	Aug. 21-27, 2022
10	Commencement of next Semester 2022-23	Sept. 1, 2022



(Dr. B. F. Jogi)

Registrar

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Date: 14-4/22

Academic calendar 2021-22 (EVEN SEMESTER)


B. Pharm. Second Year (Sem-IV)

Sr. No.	Activity	Date(s)
1	Commencement of classes	April. 18, 2022
2	Cultural Event	May. 18, 2022
3	First Sessional Examination	June. 6-11, 2022
4	Theory exam form filling for regular and supplementary examinations (without late fee)	July. 6-11, 2022
5	Exam form filling for regular and supplementary examinations (with late fee)	June. 13-17, 2022
6	End of Classes	July. 27, 2022
7	Second Sessional Examination (Theory)	July. 28-Aug. 8, 2022
8		
9	Uploading internal marks to University portal	Aug. 9-13, 2022
10	End semester theory supplementary and regular examination	Aug. 6-20, 2022
11	End Semester Practical Examination	Aug. 21-27, 2022
12	Commencement of next semester 2022-23	Sept. 1, 2022


➤ Practical sessional examination should be schedule at Institute level.

List of holidays/ festivals

Sr. No.	Holidays/ festivals	Date
1	Maharashtra Day	May. 1, 2022
2	Ramzan Eid	May. 3, 2022
3	Bodh Purnima	May. 16, 2022
4	Moharram	Aug. 9, 2022
5	Independence Day	Aug. 15, 2022
6	Ganesh Chaturthi	Aug. 31, 2022


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विद्याविहार, लोणेरे-रायगड ४०२ १०३ (महाराष्ट्र) Vidyavihar, Lonere - Raigad 402 103 (Maharashtra)

Tel: (02140) 275142 Student Helpline: 02140 - 275212

Website: www.dbatu.ac.in, E-mail: registrar@dbatu.ac.in



DBATU/Reg/AC/B. Pharma/M. Pharm /2022/77

Dated: 04/04/2022

Academic Calendar 2021-22 (EVEN SEMESTER)

B. Pharm. (Third & Final year) & M. Pharm. (Sem. - IV)

Sr. No.	Activity	Date(s)
1	Commencement of Classes	March 10, 2022
2	First Sessional Examination (Theory)	May 9-14, 2022
3	Exam form filling for Regular & Supplementary Examinations (without late fee)	June 6-11, 2022
4	Exam form filling for Regular & Supplementary Examinations (with late fee)	June 13-17, 2022
5	Second Sessional Examination (Theory)	June 23-29, 2022
6	End of Classes	July 2, 2022
7	End Semester Practical / Project Examination	July 4-9, 2022
8	Uploading Internal, Practical & Project marks to University portal	July 9-13, 2022
9	End Semester Theory Supplementary & Regular Examination	July 13-25, 2022
10	Last date of submission of M. Pharm Dissertation	August 31, 2022
11	M. Pharm final Viva-voce Examinations	September 5-10, 2022
12	Vacations	July 11-30, 2022
13	Commencement of next Semester 2022-23 for U.G. Sem. VII	Aug 1, 2022



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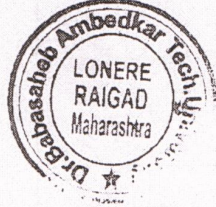
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Dr. Babasaheb Ambedkar Technological University,
LONERE 402 103,
Tal. Mangaon, Dist. Raigad, (Maharashtra)

List of Holidays

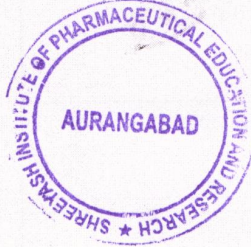
Sr. No.	Festival/Holiday	Date
1	Dhulivandan	18 th March 2022
2	GudhiPadwa	2 nd April 2022
3	Dr. BabasahebAmbedkarJayanti	14 th April 2022
4	Good Friday	15 th April 2022
5	Maharashtra Day	1 st May 2022
6	Ramzan Eid	3 rd May 2022
7	Buddha Pournima	16 th May 2022

- * All examinations will be conducted in offline mode unless otherwise notified.
- * Guidelines issued by the Government amid Covid-19 pandemic will be followed strictly.



(Dr. B F Jogi)
Registrar
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Date: -08-03-22

Academic calendar 2021-22 (EVEN SEMESTER)


B. Pharm. (Third & Final year)

Sr. No.	Activity	Date(s)
1	Commencement of classes	March. 10, 2022
2	First Sessional Examination	May. 9-14, 2022
3	Cultural Event	May. 18, 2022
4	Theory exam form filling for regular and supplementary examinations (without late fee)	June. 6-11, 2022
5	Exam form filling for regular and supplementary examinations (with late fee)	June. 13-17, 2022
6	End of Classes	July. 2, 2022
7	Second Sessional Examination (Theory)	June. 23-29, 2022
8	End Semester Practical Examination	July. 4-09, 2022
9	Uploading internal marks to University portal	July. 9-13, 2022
10	End semester theory supplementary and regular examination	July. 13-25, 2022
11	Vacations	July. 11-30, 2022
12	Commencement of next semester 2022-23	August. 1, 2022


➤ Practical sessional examination should be schedule at Institute level.

List of holidays/ festivals

Sr. No.	Holidays/ festivals	Date
1	Dhulivandan	March. 18, 2022
2	Gudhi Padwa	April. 2, 2022
3	Dr. Babasaheb Ambedkar Jayanti	April. 14, 2022
4	Good Friday	April. 15, 2022
5	Maharashtra Day	May. 1, 2022
6	Ramzan Eid	May. 3, 2022
7	Bodh Purnima	May. 16, 2022


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Vidyavihar, Lonere – Raigad 402 103 (Maharashtra)

www.dbatu.ac.in registrar@dbatu.ac.in 02140-275142

DBATU/REG/AC/B Pharm/2021/559

Date: 30/12/2021

Academic Calendar 2021-22 (B. PHARMACY)

First year B. Pharm

Sr. No.	Activity	Sem - I
1	Commencement of Classes	Dec. 13, 2021
2	First Sessional Examination (Theory)	Feb. 7-12, 2022
3	Second Sessional Examination (Theory)	March 28-April 1, 2022
4	Exam form filling	March 23-26, 2022
5	Uploading Internal Marks	April 4-8, 2022
6	End of Classes	April 9, 2022
7	End Semester Practical Examination	April 11-16, 2022
8	End Semester Theory Examination	April 18-30, 2022
9	Commencement of next Semester (Even)	May 4, 2022

Second year B. Pharmacy

Sr. No.	Activity	Sem – III (Regular)	Sem– III (DSY)
1	Commencement of Classes	Oct.4, 2021	Dec. 10, 2021
2	First Sessional Examination (Theory)	Nov. 22-25, 2022	March 2-7, 2022
3	Second Sessional Examination (Theory)	Jan. 17-21, 2022	
4	Exam form filling	Feb. 23-28, 2022	
5	Uploading Internal Marks	March 8-12, 2022	
6	End of Classes	Feb. 27, 2022	
7	End Semester Practical Examination	March 9-15, 2022	
8	End Semester Theory Examination	March 17-25, 2022	
9	Commencement of next Semester (Even)	April 1, 2022	



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List of Holidays

25 th Dec. 2021	Christmas
26 th Jan. 2022	Republic Day
19 th Feb. 2022	Chhatrapati Shivaji Maharaj Jayanti
1 st March 2022	Mahashivratri
18 th March 2022	Dhulivandan
2 nd April 2022	Gudhi Padwa
14 th April 2022	Dr. Babasaheb Ambedkar Jayanti
15 th April 2022	Good Friday
1 st May 2022	Maharashtra Day
3 rd May 2022	Ramazan Eid
16 th May 2022	Buddha Pournima



31
30.12.2021
(Dr. B. F. Jogi)
Registrar

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Hon. Vice-Chancellor Dr. B. A. Technological University, Lonere

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Academic Calendar 2021-22 (B.PHARMACY)

First year B.Pharm

Sr. No.	Activity	Date
1	Induction Program	After commencement of classes
2	Commencement of classes	Dec ,13,2021
3	First Sessional examination	February 07-12 ,2022
4	Second Sessional examination	March 28-April 1, 2022
5	Exam form filling	March 23-26,2022
6	Uploading internal marks	April 4-8, 2022
7	End of classes	April 9, 2022
8	End semester practical examination	April 11-16, 2022
9	End semester theory examination	April 18-30, 2022
10	Commencement of next semester	May 4,2022




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registrar@dbatu.ac.in

02140-275142

Registrar Office

DBATU/REG/AC/Pharmacy/2021/50:

Date: 06/08/2021

Academic Calendar 2021-22 (B. PHARMACY ODD Semester)

Sr. No.	Activity	Sem - V & VII	Sem - III
1	Commencement of Classes	Aug. 17, 2021	Sept. 27, 2021
2	First Sessional Examination (Theory)	Oct. 4-9, 2021	Nov. 22-27, 2021
3	Second Sessional Examination (Theory)	Dec. 6-11, 2021	Jan. 10-13, 2022
4	Exam form filling	Dec. 1-10, 2021	Jan. 3-8, 2022
5	Uploading Internal Marks	Dec 13-18, 2021	Jan. 14-17, 2022
6	End of Classes	Dec. 18, 2021	Jan. 15, 2022
7	End Semester Practical Examination	Dec. 20-27, 2021	Jan. 17-22, 2022
8	End Semester Theory Examination	Dec. 29- Jan 8, 2022	Jan. 25-31, 2022
9	Commencement of next Semester (Even)	Jan. 17, 2022	Feb. 7, 2022



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LONERE 402 103,
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Academic Calendar 2021-22 (B.PHARMACY ODD Semester)

Sr. No.	Activity	Sem-V & VII	Sem-III
1	Induction Program	After commencement of classes	After commencement of classes
2	Commencement of classes	August 17,2021	September 27,2021
3	First Sessional examination	October 4-9,2021	Nov. 22-27,2021
4	Second Sessional examination	December 6-11 ,2021	January 10-13 ,2022
5	Exam form filling	December 1-10 ,2021	January 3-8 ,2022
6	Uploading internal marks	December 13-18 ,2021	January 14-17,2022
7	End of classes	December 18 ,2021	January 15,2022
8	End semester practical examination	December 20-27 ,2021	January 17-22 ,2022
9	End semester theory examination	December 29-January 8 ,2022	January 25-31 ,2022
10	Commencement of next semester	January 17 ,2022	February 7,2022


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Second year B.Pharmacy

Sr. No.	Activity	Sem-III(Regular)	Sem-III(DSY)
1	Commencement of classes	Oct 4,2021	Dec 10,2021
2	First Sessional examination	Nov 22-25,2022	March 2-7,2022
3	Second Sessional examination	Jan 17-21,2022	
4	Exam form filling	Feb 23-28,2022	
5	Uploading internal marks	March 18-12,2022	
6	End of classes	Feb 27,2022	
7	End semester practical examination	March 09-15,2022	
8	End semester theory examination	March 17-25, 2022	
9	Commencement of next semester	April 1,2022	

25 Dec,2021	Christmas
26 Jan, 2022	Republic day
19 Feb, 2022	Chhatrapati Shivaji Maharaj Jayanti
1March,2022	Mahashivratri
18 March 2022	Dhulivandan
2 April 2022	Gudhipadwa
14 April 2022	Dr.Babasaheb Ambedkar Jayanti
15 April 2022	Good friday
1 May 2022	Maharashtra day
3 May 2022	Ramazan Eid
16 May 2022	Buddha Pournima


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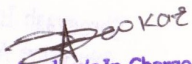


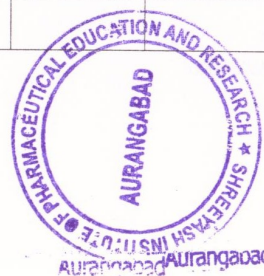

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WORK LOAD 2021-2022 (EVEN SEMESTER)


Sr No.	Name of faculty	F.Y.B. PHARM SEM II		S.Y.B. PHARM SEM III		S.Y.B. PHARM SEM IV		T.Y.B. PHARM SEM VI		FINAL YEAR SEM VIII		TOTAL
		Theory	Practical	Theory	Practical	Theory	Practical	Theory	Practical	Theory	Practical	
1	Dr. Ganesh Tapdiya											
2	Mr. Milind D Kamble									Biostat (6+2)		6+2
3	Ms. Rashmi Tambare						PP II (4x2=08)	QA (6+2=08)				14+2
4	Ms. Shradhha Nalawade					PP-I I (6+2)	PP II (4x3=12)					18+2
5	Mrs. Snehal Pawar	HAP-II (6+2=8)	HAP-II (4x3=12)									18+2
6	Mrs. Neha Gandhi							BP & PK (6+2)	HDT (4x2=8)			14+2
7	Mrs. Priti Undre					Medichem I (6+2)	Medichem I (4x3=12)					18+2
8	Mr. Tushar Joshi						P'col I (4x1=04)		Mediche-III (4x2=8)			10+2
9	Ms. Vishakha Shelke	POC-I (6+2=8)	POC-I (4x3=12)									18+2
10	Mr. Sayeed Shaikh		POC-I (4x2=08)							AIT (6+2)		14+2
11	Mr. Sachin Dhawale					POC-III (6+2=8)	Medichem I (4x2=8)					14+2
12	Mrs. Momin Aaliya							P'col III (6+2=8)	P'col III (4x3=12)			18+2
13	Mrs. Sneha Khandale				PE (4x1=04)					P'covigilance (6+2=8)		10+2
14	Mrs. Pallavi Bhosale	PATHO (6+2=8)	HAP-II (4x2=8)									14+2


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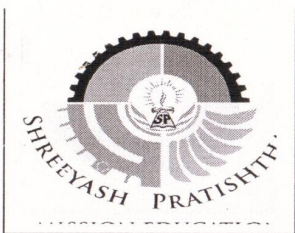

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15	Mrs. Aditi Jyotishi							Biotech (6+2=8)	HDT (4x3=12)			18+2
16	Mr. Khalse Rupesh							HDT (6+2=8)	P ^o col III (4x2=8)			14+2
17	Ms. Shubhangi Deshmukh	EVS (4+2)										4+2
18	Mrs. Minal Chaudhary					P ^o col I (6+2=8)	P ^o col I (4x3=12)					18+2
19	Mrs. Geetanjali Patil							Medichem III (6+2=8)	Medichem III (4x3=12)			18+2
20	Mr. Vinayak Mhaismale					P ^o cog I (4x2=8)	P ^o cog I (4x3=12)					18+2
21	Mr. Pathan Hujeb						P ^o col-I (4x1=04), P ^o cog I (4x2=08)					10+2
22	Arundhati Dekar											00
23	Mr. Kishan Jaiswal	CA (4+2=06)	CA (2x6=12)									16+2
24	Hrshikesh Rajput	BIOCHEM (4x3=12)				PE (4x1=04)						14+2
25	Siddhart Chatse				PE (3+1=04)					SPP(6+2=08)		10+2
26	Dr. Mangesh Ghodke	BIOCHEM (4x2=8)	BIOCHEM (4x2=8)									14+2

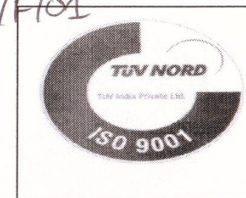

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FORMAT NO: STC/PR.08/F/01



SHREEYASH PRATISHTHAN

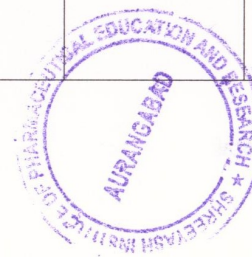
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SHREEYASH INSTITUTE OF PHARMACEUTICAL EDUCATION AND RESEARCH

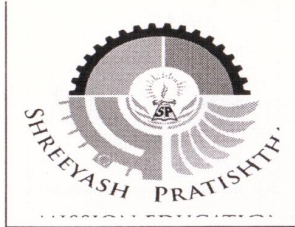
WORK LOAD 2021-2022 (ODD SEMISTER)

No.	Name of faculty	F.Y.B. PHARM SEM I		S.Y.B.PHARM SEM IV		T.Y.B.PHARM SEM V		FINAL YEAR SEM VII		TOTAL
		Theory	Practical	Theory	Practical	Theory	Practical	Theory	Practical	
1	Dr. Ganesh Tapdiya									
2	Mr. Milind D Kambale							IP -II (6+2)=08	IMA (4X2)=08	14+2 hr
3	Ms. Shradhha Nalawade					IP-I [6+2]	IP -I 4X3=12			18+2
4	Mr. Mohammed Sayyed							IMA (6+2)	IMA 4X3=12	18+2
7	Mr. Mahendra Khandare			POC-II (6+2)=08	POC-II (4X3)=12					18+2
8	Ms. Vishakha Shelke					M.Che -II (6+2)=8	P'cog- II (4 X 2=8)			14+2
9	Mrs.PritiUndre			PE 6+2=8	PE 4X3=12					18+2
10	Mr. Sachin Dhawale					PJ (6+2)=8				08
11	Mrs. Sneha Khandale				P'Micro (4X2=8)			PP (6+2)		14+2

[Signature]
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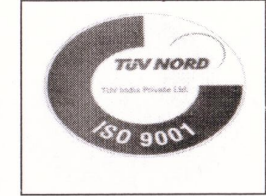
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SHREEYASH INSTITUTE OF PHARMACEUTICAL EDUCATION AND RESEARCH

(2021-22) odd semester



12	Mrs. Pallavi Bhosale					P'col -II (6+2)=8	P'col -II 4X3=12			18+2
13	Mrs. Minal Chaudhary			PP- I (6+2)=08	PP-I (4X3)=12					18+2
14	Mrs. Arundhati Deokar					P' cog II (6+2)=8	P' cog II (4X3)=12			18+2
15	Mrs. Shubhangi Deshmukh						IP -I 4X2=08			08
16	Vinayak Mhaismale				PP-I (4X2)=8			NDDS (6+2)=8		14+2
17	Aditi Jyotishi				POC-II (4X2)=08					08
19	Tushar Joshi			P'Micro (6+2)=8	P'Micro (4X3)=12					14+2
20	Rupesh Khalse				PE 4X2=8		P'col -II 4X2=08			14+2
21	Mrs. Momin Aaliya									
22	Mr. Kishan Jaiswal									
23	Mrs. Rashmi Tambe									

A. Deokar
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SHREEYASH INSTITUTE OF PHARMACEUTICAL EDUCATION AND RESEARCH

WORK LOAD ~~2019-20~~ (ODD SEMESTER)




(2021-22)

Sr. No.	Name of faculty	F.Y.B. PHARM SEM I		TOTAL
		Theory	Practical	
1	Dr. Ganesh Tapdiya			
2	Mr. Milind D Kamble		PCEU Lab 4X2=08	08hr
3	Ms. Snehal Pawar	PCEU (6+2)=08	PCEU Lab 4X3=12	20
4	Mr. Mohammed Sayeed		PIC Lab 4X2=08	08
5	Mr. Sneha Lande	RM (2+2)=4		4
6	Ms. Vishakha Shelke	PA (6+2)=8	PA Lab 4X3=12	20
7	Mr. Sachin Dhawale	PIC (6+2)=8	PIC Lab 4X3=12	20
8	Mrs. Sneha Khandale		HAP Lab 4X2=08	08
9	Mrs. Pallavi Bhosale	HAP-1 (6+2)=8	HAP-1 Lab 4X3=12	20
10	Mrs. Minal Chaudhary		PA Lab 4X2=08	08
11	Mr. Kishan Jaiswal	CSK (2+2)=4	CSK Lab 5X2=10	14


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DEPARTMENT OF PHARMACY
CLASS TIME TABLE (DIVISION-B)

Class: B. Pharmacy First year (Second Semester)

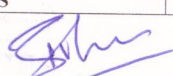
Academic Year 2021-2022

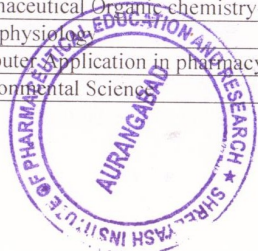
w.e.f.: 01 / 06/2022

TIME	Batch	(MORNING) PRACTICAL	01:00 pm to 01:45 pm	(AFTERNOON) THEORY			
		09:00am-1:00 pm		01:45-02:45 pm	02:45-03:45 pm	03:45-04:45 pm	04:45-05:45 pm
DAY	Batch	Subject	R E C E S S				
MON	E	Biochem [MSG](211)		Biochem [MSG] (314)	EVS[SD] (314)	HAP-II [SP](314)	Biochem [MSG] Tutorial (314)
TUE	D E	Biochem [MSG](211) CA[KBJ](203)		HAP-II [SP](314)	Biochem [MSG] (314)	CA[KBJ] (314)	HAP-II[SP] Tutorial (314)
WED	D E	CA[KBJ](203) POC-I [SMS](209)		PTPH [PVB] (314)	HAP-II [SP](314)	Biochem [MSG] (314)	PTPH[PVB] Tutorial (314)
THU	D	POC-I [SMS](209)		CONTACT/ MEETING HOUR			
FRI	D E	HAP-II[PVB](312) ↔		POC-I[VRS] (314)	PTPH [PVB] (314)	POC-I[VRS] (314)	EVS[SD] (314)
SAT	D E	↔ HAP-II[PVB](312)		CA[KBJ] (314)	POC-I[VRS] (314)	PTPH [PVB] (314)	POC-I[VRS] Tutorial (314)


SUBJECTS	
Abbr.	Full name
Biochem	Biochemistry
HAP-II	Human Anatomy and Physiology-II
POC-I	Pharmaceutical Organic chemistry-I
PTPH	Pathophysiology
CA	Computer Application in pharmacy
EVS	Environmental Science

FACULTY	
Abbr.	Full name
MSG	Mangesh S Ghodke
SP/PVB	Snehal Pawar/ Pallavi Bhosle
VRS/SMS	Vishakha Shelke/ Mohammad Sayyed
PVB	Pallavi Bhosle
KBJ	Kishan Jaiswal
SD	Shubhangi Deshmukh


CLASS INCHARGE
Pallavi Bhosle




MASTER TIME TABLE INCHARGE
Momin Aaliya Firdus


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DEPARTMENT OF PHARMACY
CLASS TIME TABLE (DIVISION-A)

Class: B. Pharmacy First year (Second Semester)

Academic Year 2021-2022

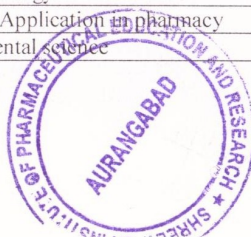
w.e.f.: 01/06/2022

TIME	Batch	(MORNING) PRACTICAL	R E C E S S	(AFTERNOON) THEORY			
		09:00am-1:00 pm		01:00 pm to 01:45 pm	01:45-02:45 pm	02:45-03:45 pm	03:45-04:45 pm
DAY	Batch	Subject					
MON	A B C	HAP-II[SP](312) CA[KBJ](203) ↔		POC-I [VRS](301)	PTPH [PVB] (301)	EVS[SD] (301)	PTPH [PVB] Tutorial (301)
TUE	A B C	POC-I[VRS](209) HAP-II[SP](312) ↔		PTPH [PVB] (301)	POC-I [VRS](301)	PTPH [PVB] (301)	POC-I [VRS] Tutorial (301)
WED	A B C	↔ Biochem [HR](211) HAP-II[SP](312)		Biochem [MSG] (301)	CA[KBJ] (301)	POC-I [VRS](301)	HAP-II [SP] Tutorial (301)
THU	A B C	↔ ↔ CA[KBJ](203)		CONTACT/MEETING HOUR			
FRI	A B C	Biochem [HR](211) ↔ POC-I[VRS](209)		Biochem [MSG] (301)	CA[KBJ] (301)	HAP-II [SP](301)	Biochem [MSG] Tutorial (301)
SAT	A B C	CA[KBJ](203) POC-I[VRS](209) Biochem[HR](211)		HAP-II [SP](301)	Biochem [MSG] (301)	EVS[SD] (301)	HAP-II [SP](301)

SUBJECTS	
Abbr.	Full name
Biochem	Biochemistry
HAP-II	Human Anatomy and Physiology-II
POC-I	Pharmaceutical Organic chemistry-I
PTPH	Pathophysiology
CA	Computer Application in pharmacy
EVS	Environmental Science

FACULTY	
Abbr.	Full name
MSG/HR	Mangesh S Ghodke /Hrishikesh Rajput
SP	Snehal Pawar
VRS	Vishakha Shelke
PVB	Pallavi Bhosle
KBJ	Kishan Jaiswal
SD	Shubhangi Deshmukh

CLASS INCHARGE
Pallavi Bhosle



MASTER TIME TABLE INCHARGE
Momin Aaliya Firdous

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FORMAT NO. STC/PR08/F/03



DEPARTMENT OF PHARMACY
CLASS TIME TABLE (DIVISION-A)

Class: B. Pharmacy Second year (Fourth Semester)

Academic Year 2021-2022

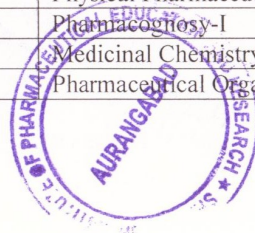
w.e.f.: 18 /04 /2022

TIME	(MORNING)THEORY				(AFTERNOON) PRACTICAL	
	9:00 -10:00am	10-11:15 am	11:15-12:15 pm	12:15-1:15 pm	01:15-2:00pm	2:00-6:00 pm
DAY					Batch	Subject
MON	PP-II(SVN) Tutorial (401)	P'COLOGY-I [MYC](401)	PP-II(SVN) (401)	P'COLOGY-I[MYC] Tutorial (401)	R E C E S S	A P'COG-I [HPK](310) B M.CHEM-I [PU](211) C PP-II (SVN)(013)
TUE	POC-III[SD] Tutorial(401)	PP-II(SVN) (401)	POC-III[SD] (401)	P'COLOGY-I [MYC](401)		A P'COLOGY-I[MYC](414) B P'COG-I [HPK](310) C M.CHEM-I [PU](211)
WED	LIBRARY	POC-III[SD] (401)	P'COLOGY-I [MYC](401)	PP-II(SVN) (401)		A M.CHEM-I [PU](211) B ←→ C P'COG-I [VM](310)
THU	LIBRARY	M.CHEM-I [PU](401)	P'COG-I [VM](401)	M.CHEM-I[PU] Tutorial(401)		CONTACT/MEETING HOUR
FRI	ASSIGNMENT	P'COG-I [VM](401)	M.CHEM-I [PU](401)	P'COG-I[VM] Tutorial(401)		A ←→ B PP-II (SVN)(013) C P'COLOGY-I[MYC](414)
SAT	ASSIGNMENT	POC-III[SD] (401)	P'COGI[VM] (401)	M.CHEM-I[PU] (401)		A PP-II (SVN)(013) B P'COLOGY-I[MYC](414) C ←→

SUBJECTS	
Abbr.	Full Name
P'COLOGY-I	Pharmacology-I
PP-II	Physical Pharmaceutics-II
P'COG-I	Pharmacognosy-I
M.CHEM-I	Medicinal Chemistry-I
POC-III	Pharmaceutical Organic Chemistry-III

FACULTY	
Abbr.	Full Name
MYC	Minal Chaudhari
SVN	Shraddha Nalawade
VM/HPK	Vinayak Mhaismale/ Hujeb Pathan Khan
PU	Priti Undre
SD	Sachin Dhawale

Undre
CLASS INCHARGE
Priti undre



Aaliya
MASTER TIME TABLE INCHARGE
Momin Aaliya Firdous

SDP
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DEPARTMENT OF PHARMACY
CLASS TIME TABLE (DIVISION-B)

Class: B. Pharmacy Second year (Fourth Semester)

Academic Year 2021-2022

w.e.f: 18 / 04 /2022

TIME	(MORNING)THEORY				(AFTERNOON) PRACTICAL		
	9:00 -10:00am	10-11:15 am	11:15-12:15 pm	12:15-1:15 pm	01:15-2:00pm	2:00-6:00 pm	
DAY					Batch	Subject	
MON	LIBRARY	M.CHEM-I[PU](402)	P'COG-I [VM](402)	POC-III[SD](402)	R E C E S S	D E P'COLOGY-I[TJ](414)	
TUE	ASSIGNMENT	P'COG-I [VM](402)	M.CHEM-I [PU](402)	P'COG-I[VM] Tutorial(402)		D E PP-II (RT)(013)	
WED	ASSIGNMENT	M.CHEM-I [PU](402)	P'COG-I [VM](402)	M.CHEM-I[PU] Tutorial(402)		D E PP-II (RT)(013) P'COLOGY-I[HPK](414)	
THU	P'COLOGY-I Tutorial[MYC] (402)	P'COLOGY-I [MYC](402)	PP-II(SVN) (402)	POC-III[SD] (402)		CONTACT/MEETING HOUR	
FRI	PP-II(SVN) Tutorial (402)	POC-III[SD] (402)	PP-II(SVN) (402)	P'COLOGY-I [MYC](402)		D E M.CHEM-I [SD](211) P'COG-I [VM](310)	
SAT	UNIT TEST	PP-II(SVN) (402)	P'COLOGY-I [MYC](402)	POC-III[SD] Tutorial (402)		D E P'COG-I [VM](310) M.CHEM-I [SD](211)	

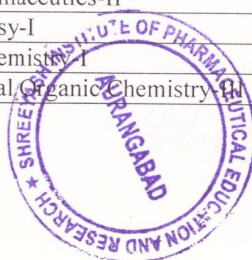
SUBJECTS	
Abbr.	Full Name
P'COLOGY-I	Pharmacology-I
PP-II	Physical Pharmaceutics-II
P'COG-I	Pharmacognosy-I
M.CHEM-I	Medicinal Chemistry-I
POC-III	Pharmaceutical Organic Chemistry

FACULTY	
Abbr.	Full Name
MYC/HPK/TJ	Minal Chaudhari/Hujeb Pathan /Tushar Joshi
SVN/RT	Shraddha Nalawade/Rashmi Tambare
VM	Vinayak Mhaismale
PU/SD	Priti Undre/ Sachin Dhawale
SD	Sachin Dhawale

CLASS INCHARGE
Priti undre

MASTER TIME TABLE INCHARGE
Momin Aaliya Firdous

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DEPARTMENT OF PHARMACY
CLASS TIME TABLE (DIVISION-A)

Class: B. Pharmacy Third year (Six Semester)

Academic Year 2021-2022

w.e.f.: 10 /03 /2022

TIME	(MORNING)THEORY				(AFTERNOON) PRACTICAL		
	9:00 -10:00am	10-11:15 am	11:15-12:15 pm	12:15-1:15 pm	01:15-2:00pm	2:00-6:00 pm	
DAY					R E C E S S	Batch Subject	
MON	QA(RT) (301) Tutorial	M.CHEM-III(GGP)(301)	P'COLOGY-III [MAF](301)	BPPK(NG)(301)		A B C	P'COLOGY-III[VM](312) M.CHEM-III(GGP)[209]
TUE	M.CHEM-III (GGP)(301) Tutorial	HDT [RBK] (301)	BPPK(NG) (301)	M.CHEM-III (GGP)(301)		A B C	P'COLOGY-III[MAF](312)
WED	BPPK(NG) (301) Tutorial	P'BIOTECH (VM)(301)	HDT[RBK] (301)	M.CHEM-III (GGP) (301)		A B C	P'COLOGY-III[VM](312) HDT [RBK](310)
THU	P'BIOTECH(VM) (301)Tutorial	QA(RT) (301)	P'COLOGY-III [MAF] (301)	QA(RT) (301)			CONTACT HOUR /MEETING
FRI	P'COLOGY-III(301) [MAF] Tutorial	QA(RT) (301)	HDT[RBK] (301)	P'BIOTECH (VM)(301)		A B C	M.CHEM-III(GGP)[209] HDT [RBK](310)
SAT	HDT [RBK] (301)	P'BIOTECH(VM) (301)	BPPK(NG) (301)	P'COLOGY-III [MAF](301)		A B C	HDT [RBK](310) M.CHEM-III(GGP)[209]

SUBJECTS	
Abbr.	Full Name
BPPK	Biopharmaceutics and Pharmacokinetics
HDT	Herbal Drug Technology
P'COLOGY-III	Pharmacology-III
M.CHEM-III	Medicinal Chemistry-III
P'BIOTECH	Pharmaceutical Biotechnology
QA	Quality Assurance

FACULTY	
Abbr.	Full Name
NG	Dr.Neha Gandhi
RBK/NG	R.B Khalse
MAF/VM	Momin Aaliya / Vinayak Mhaismale
GGP	Geetanjali Patil
VM	Vinayak Mhaismale
RT	Rashmi Tambare

CLASS INCHARGE
Geetanjali Patil

MASTER TIME TABLE INCHARGE
Momin Aaliya Firdous

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SHREEYASH INSTITUTE OF PHARMACEUTICAL EDUCATION & RESEARCH

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DEPARTMENT OF PHARMACY
CLASS TIME TABLE (DIVISION-B)

Class: B. Pharmacy Third year (Six Semester)

Academic Year 2021-2022

w.e.f.:10 /03 /2022

TIME	(MORNING)THEORY				(AFTERNOON) PRACTICAL		
	9:00 -10:00am	10-11:15 am	11:15-12:15 pm	12:15-1:15 pm	01:15-2:00pm	2:00-6:00 pm	
DAY					Batch	Subject	
MON	P'BIOTECH(VM) (314) Tutorial	HDT [RBK](314)	BPPK(NG) (301)	P'COLOGY-III [MAF](314)	R E C E S S	↔ HDT [NG](310)	
TUE	QA(RT) (314) Tutorial	P'COLOGY-III [MAF](314)	HDT[RBK] (314)	BPPK(NG) (314)		D E	HDT [NG](310) M.CHEM-III(HR)[209]
WED	P'COLOGY-III Tutoria[MAF](314)	HDT[RBK] (314)	QA(RT) (314)	BPPK(NG) (314)		D E	M.CHEM-III(HR)[209] ↔
THU	M.CHEM-III (GGP)(314) Tutorial	P'BIOTECH(VM) (314)	M.CHEM-III (GGP) (314)	P'COLOGY-III [MAF](314)		CONTACT HOUR /MEETING	
FRI	HDT [RBK] (314) Tutorial	P'BIOTECH(VM) (314)	M.CHEM-III (GGP)(314)	QA(RT) (314)		D E	↔ P'COLOGY-III[MAF](312)
SAT	BPPK (NG) Tutorial(314)	QA(RT) (314)	M.CHEM-III (GGP) (314)	P'BIOTECH(VM) (314)		D E	↔ P'COLOGY-III[MAF](312)

SUBJECTS	
Abbr.	Full Name
BPPK	Biopharmaceutics and Pharmacokinetics
HDT	Herbal Drug Technology
P'COLOGY-III	Pharmacology-III
M.CHEM-III	Medicinal Chemistry-III
P'BIOTECH	Pharmaceutical Biotechnology
QA	Quality Assurance

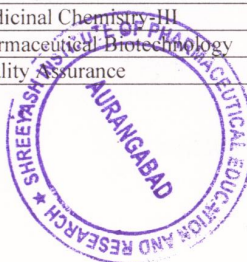
FACULTY	
Abbr.	Full Name
NG	Dr.Neha Gandhi
RBK/SK/NG	R.B Khalse/Neha Gandhi
MAF	Momin Aaliya Firdous
GGP/HR	Geetanjali Patil/Hrishikesh Rajput
VM	Vinayak Mhaismale
RT	Rashmi Tambare

CLASS INCHARGE
Geetanjali Patil

MASTER TIME TABLE INCHARGE
Momin Aaliya Firdous

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Education And Research, Aurangabad





SHREEYASH INSTITUTE OF PHARMACEUTICAL EDUCATION & RESEARCH

DEPARTMENT OF PHARMACY
CLASS TIME TABLE (DIVISION-A)

Class: B. Pharmacy Third year (Six Semester)

Academic Year 2021-2022

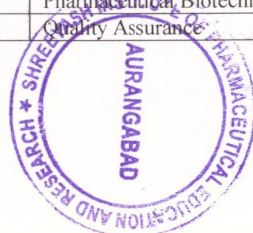
w.e.f.: 18/04 /2022

TIME	(MORNING)THEORY				(AFTERNOON) PRACTICAL	
	9:00 -10:00am	10-11:15 am	11:15-12:15 pm	12:15-1:15 pm	01:15-2:00pm	2:00-6:00 pm
DAY					Batch	Subject
MON	QA(RT) (301) Tutorial	M.CHEM-III(GGP)(301)	P*COLOGY-III [MAF](301)	BPPK(NG) (301)	A B C	P*COLOGY-III[RBK](213) M.CHEM-III(GGP)[209]
TUE	M.CHEM-III (GGP)(301) Tutorial	HDT [RBK] (301)	M.CHEM-III (GGP)(301)	BPPK(NG) (301)	A B C	←→ P*COLOGY-III[MAF](213)
WED	BPPK(NG) (301) Tutorial	P*BIOTECH (AJ)(301)	BPPK(NG) (301)	M.CHEM-III (GGP) (301)	A B C	←→ P*COLOGY-III[RBK](213) HDT [AJ](312)
THU	P*BIOTECH(AJ) (301)Tutorial	QA(RT) (301)	P*COLOGY-III [MAF] (301)	QA(RT) (301)	CONTACT HOUR /MEETING	
FRI	P*COLOGY-III(301) [MAF] Tutorial	QA(RT) (301)	HDT[RBK] (301)	P*BIOTECH (AJ) (301)	A B C	M.CHEM-III(GGP)[209] HDT [AJ](312) -----
SAT	HDT [RBK] (301) Tutorial	P*BIOTECH(AJ) (301)	HDT[RBK] (301)	P*COLOGY-III [MAF](301)	A B C	HDT [AJ](312) ----- M.CHEM-III(GGP)[209]

SUBJECTS	
Abbr.	Full Name
BPPK	Biopharmaceutics and Pharmacokinetics
HDT	Herbal Drug Technology
P*COLOGY-III	Pharmacology-III
M.CHEM-III	Medicinal Chemistry-III
P*BIOTECH	Pharmaceutical Biotechnology
QA	Quality Assurance

FACULTY	
Abbr.	Full Name
NG	Dr.Neha Gandhi
RBK/AJ	R.B Khalse/Aditi Jyotishi
MAF/RBK	Momin Aaliya /RB Khalse
GGP	Geetangali Patil
AJ	Aditi Jyotishi
RT	Rashmi Tambare

CLASS INCHARGE
Geetanjali Patil



MASTER TIME TABLE INCHARGE
Momin Aaliya Firdous

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SHREEYASH INSTITUTE OF PHARMACEUTICAL EDUCATION & RESEARCH

**DEPARTMENT OF PHARMACY
CLASS TIME TABLE (DIVISION-B)**

Class: B. Pharmacy Third year (Six Semester)

Academic Year 2021-2022

w.e.f.:18 /04 /2022

TIME	(MORNING)THEORY				(AFTERNOON) PRACTICAL		
	9:00 -10:00am	10-11:15 am	11:15-12:15 pm	12:15-1:15 pm	01:15-2:00pm	2:00-6:00 pm	
DAY					R E C E S S	Batch	Subject
MON	P'BIOTECH(AJ) (314) Tutorial	HDT [RBK](314)	BPPK(NG) (301)	P'COLOGY-III [MAF](314)		D E	←————→ HDT [NG] (310)
TUE	QA(RT) (314) Tutorial	P'COLOGY-III [MAF](314)	BPPK(NG) (314)	P'BIOTECH(AJ) (314)		D E	HDT [NG](310) M.CHEM-III(TJ)[209]
WED	P'COLOGY-III Tutorial[MAF(314)	HDT[RBK] (314)	QA(RT) (314)	BPPK(NG) (314)		D E	M.CHEM-III(TJ)[209] ←————→
THU	M.CHEM-III (GGP)(314) Tutorial	HDT[RBK] (314)	M.CHEM-III (GGP) (314)	P'COLOGY-III [MAF](314)		CONTACT HOUR /MEETING	
FRI	HDT [RBK] (314) Tutorial	P'BIOTECH(AJ) (314)	M.CHEM-III (GGP)(314)	QA(RT) (314)		D E	←————→ P'COLOGY-III[MAF](213)
SAT	BPPK (NG) Tutorial(314)	QA(RT) (314)	M.CHEM-III (GGP) (314)	P'BIOTECH(AJ) (314)		D E	P'COLOGY-III[MAF](213) ←————→

SUBJECTS	
Abbr.	Full Name
BPPK	Biopharmaceutics and Pharmacokinetics
HDT	Herbal Drug Technology
P'COLOGY-III	Pharmacology-III
M.CHEM-III	Medicinal Chemistry-III
P'BIOTECH	Pharmaceutical Biotechnology
QA	Quality Assurance

FACULTY	
Abbr.	Full Name
NG	Dr.Neha Gandhi
RBK/NG	R.B Khalse/Neha Gandhi
MAF	Momin Aaliya Firdous
GGP	Geetangali Patil/Tushar Joshi
AJ	Aditi Jyotishi
RT	Rashmi Tambare

CLASS INCHARGE
Geetanjali patil

MASTER TIME TABLE INCHARGE
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FORMAT NO. STC/PR08/F/03



DEPARTMENT OF PHARMACY
CLASS TIME TABLE (DIVISION-A)

Class: B. Pharmacy Final year (Eight Semester)

Academic Year 2021-2022

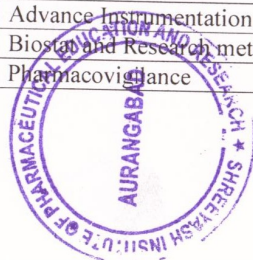
w.e.f.: 10 /03 /2022

	(MORNING) PRACTICAL		(AFTERNOON) THEORY			
TIME	09:00am-1:00 pm	01:00 pm to 01:45 pm	01:45-02:45 pm	02:45-03:45 pm	03:45-04:45 pm	04:45-05:45 pm
DAY	Subject					
MON	↔	R E C E S S	BIOSTAT&RM (MDK)(401)	AIT(SMS) (401)	ASSIGNMENT	BIOSTAT&RM (MDK)(401)
TUE	↔		AIT(SMS) (401)	BIOSTAT&RM (MDK)(401)	AIT(SMS) Tutorial(401)	LIBRARY
WED	↔		AIT(SMS) (401)	BIOSTAT&RM (MDK)(401) Tutorial	SPP(SVN) (401)	PV(SK)(401)
THU	↔		MEETING HOUR			
FRI	↔		PV(SK)(401)	SPP(SVN) (401)	LIBRARY	PV(SK)(401) Tutorial
SAT	↔		SPP(SVN) (401)	PV(SK)(401)	SPP(SVN) Tutorial(401)	ASSIGNMENT

SUBJECTS	
Abbr.	Full name
SPP	Social and Preventive Pharmacy
AIT	Advance Instrumentation Technology
BIOSTAT&RM	Bio-stat and Research methodology
PV	Pharmacovigilance

FACULTY	
Abbr.	Full name
SVN	Shraddha Nalawade
MS	Mohamed Sayyed
MDK	Milind Kamble
SK	Sneha Khandale

Sneha
 CLASS INCHARGE
 Sneha Khandale



Momin
 MASTER TIME TABLE INCHARGE
 Momin Aaliya Firdous

Sneha
 PRINCIPAL
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FORMAT NO. STC/PR08/F/03



DEPARTMENT OF PHARMACY
CLASS TIME TABLE (DIVISION-B)

Class: B. Pharmacy Final year (Eight Semester)

Academic Year 2021-2022

w.e.f.: 10 /03 /2022

	(MORNING) PRACTICAL		(AFTERNOON) THEORY			
TIME	09:00am-1:00 pm	01:00 pm to 01:45 pm	01:45-02:45 pm	02:45-03:45pm	03:45-04:45 pm	04:45-05:45 pm
DAY	Subject					
MON	↔	R E C E S S	PV(SK)(402)	SPP(SVN)(402)	PV(SK)(402)	SPP(SVN)(402)
TUE	↔		SPP(SVN)(402)	ASSIGNMENT	PV(SK)(402)	SPP(SVN) Tutorial(402)
WED	↔		BIOSTAT&RM (MDK)(402)	PV(SK)(402) Tutorial	LIBRARY	AIT(MS) Tutorial(402)
THU	↔		MEETING HOUR			
FRI	↔		LIBRARY	AIT(MS)(402)	ASSIGNMENT	BIOSTAT&RM (MDK)(402)
SAT	↔		AIT(MS)(402)	BIOSTAT&RM (MDK)(402)	AIT(MS)(402)	BIOSTAT&RM Tutorial(MDK)(402)

SUBJECTS	
Abbr.	Full name
SPP	Social and Preventive Pharmacy
AIT	Advance Instrumentation Technology
BIOSTAT&RM	Biostat and Research Methodology
PV	Pharmacovigilance

FACULTY	
Abbr.	Full name
SVN	Shraddha Nalawade
MS	Mohamed Sayyed
MDK	Milind Kamble
SK	Sneha Khandale

Sneha Khandale
 CLASS INCHARGE
 Sneha Khandale



Momin Aaliya Firdous
 MASTER TIME TABLE INCHARGE
 Momin Aaliya Firdous

Sneha Khandale
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**DEPARTMENT OF PHARMACY
CLASS TIME TABLE (DIVISION-A)**

Class: B. Pharmacy Final year (Eight Semester)

Academic Year 2021-2022

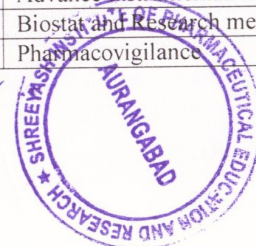
w.e.f.: 18 /04 /2022

	(MORNING) PRACTICAL		(AFTERNOON) THEORY			
TIME	09:00am-1:00 pm	01:00 pm to 01:45 pm	01:45-02:45 pm	02:45-03:45 pm	03:45-04:45 pm	04:45-05:45 pm
DAY	Subject					
MON	↔	R E C E S S	BIOSTAT&RM (MDK)(401)	AIT(SMS) (401)	ASSIGNMENT	BIOSTAT&RM (MDK)(401)
TUE	↔		AIT(SMS) (401)	BIOSTAT&RM (MDK)(401)	AIT(SMS) Tutorial(401)	LIBRARY
WED	↔		AIT(SMS) (401)	BIOSTAT&RM (MDK)(401)Tutorial	ASSIGNMENT	PV(SK)(401)
THU	↔		MEETING HOUR			
FRI	↔		PV(SK)(401)	SPP(SC) (401)	PV(SK) Tutorial (401)	SPP(SC) (401)
SAT	↔		SPP(SC) (401)	PV(SK)(401)	SPP(SC) Tutorial(401)	ASSIGNMENT

SUBJECTS	
Abbr.	Full name
SPP	Social and Preventive Pharmacy
AIT	Advance Instrumentation Technology
BIOSTAT&RM	Biostat and Research methodology
PV	Pharmacovigilance

FACULTY	
Abbr.	Full name
SC	Siddhart Chatse
MS	Mohamed Sayyed
MDK	Milind Kamble
SK	Sneha Khandale

Sneha Khandale
CLASS INCHARGE
Sneha Khandale



Momin Aaliya Firdous
MASTER TIME TABLE INCHARGE
Momin Aaliya Firdous

Sneha Khandale
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DEPARTMENT OF PHARMACY
CLASS TIME TABLE (DIVISION-B)

Class: B. Pharmacy Final year (Eight Semester)

Academic Year 2021-2022

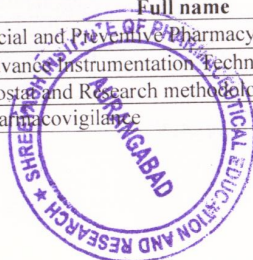
w.e.f.: 18 /04 /2022

	(MORNING) PRACTICAL		(AFTERNOON) THEORY			
TIME	09:00am-1:00 pm	01:00 pm to 01:45 pm	01:45-02:45 pm	02:45-03:45pm	03:45-04:45 pm	04:45-05:45 pm
DAY	Subject	R E C E S				
MON	↔		PV(SK)(402)	SPP(SC) (402)	PV(SK)(402)	SPP(SC) (402)
TUE	↔		SPP(SC) (402)	ASSIGNMENT	PV(SK)(402)	SPP(SC) Tutorial(402)
WED	↔		BIOSTAT&RM (MDK)(402)	PV(SK)(402) Tutorial	LIBRARY	AIT(MS) Tutorial(402)
THU	↔		MEETING HOUR			
FRI	↔		BIOSTAT&RM (MDK)(402)	ASSIGNMENT	AIT(MS) (402)	LIBRARY
SAT	↔		AIT(MS) (402)	BIOSTAT&RM (MDK)(402)	AIT(MS) (402)	BIOSTAT&RM Tutorial(MDK) (402)

SUBJECTS	
Abbr.	Full name
SPP	Social and Preventive Pharmacy
AIT	Advanced Instrumentation Technology
BIOSTAT&RM	Biostat and Research methodology
PV	Pharmacovigilance

FACULTY	
Abbr.	Full name
SC	Siddhart Chatse
MS	Mohamed Sayyed
MDK	Milind Kamble
SK	Sneha Khandale

CLASS INCHARGE
Sneha Khandale



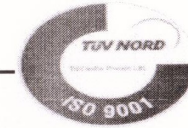
MASTER TIME TABLE INCHARGE
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FORMAT NO. STC/PR08/F/03



DEPARTMENT OF PHARMACY
CLASS TIME TABLE (DIVISION-A)

Class: B. Pharmacy First year (First Semester)

Academic Year 2021-2022

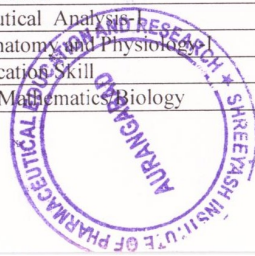
w.e.f.: 5/1/2022

TIME	Batch	(MORNING) PRACTICAL	R E C E S S	(AFTERNOON) THEORY			
		09:00am-1:00 pm		01:00 pm to 01:45 pm	01:45-02:45 pm	02:45-03:45 pm	03:45-04:45 pm
DAY	Batch	Subject					
MON	A	HAP-I[PVB](312)		HAP-I[PVB](301)	PA-I [VRS](301)	HAP-I[PVB](301)	CS[KBJ](301)
	B	PCEUTICS-I[SP](013)					
TUE	A	PA-I[VRS](211)		PA-I Tutorial [VRS](301)	HAP-I [PVB](301)	PA-I [VRS](301)	HAP-I [PVB] Tutorial(301)
	B	HAP-I[PVB](312)					
	C	PCEUTICS-I[SP](013)					
WED	B	PIC [SD](209)		CS[KBJ](301)	PA-I [VRS](301)	PIC [SD] Tutorial (301)	PCEUTICS-I Tutorial [SP](301)
	C	HAP-I[PVB](312)					
THU	A	PCEUTICS-I[SP](013)		CONTACT/MEETING HOUR			
	C	CS[KBJ](203)					
FRI	A	PIC [SD](209)	PIC [SD](301)	PCEUTICS-I [SP](301)	PIC [SD](301)	RM[SL](301)	
	B	CS[KBJ](203)					
	C	PA-I[VRS](211)					
SAT	A	CS[KBJ](203)	PCEUTICS-I [SP](301)	PIC [SD](301)	RM[SL](301)	PCEUTICS-I [SP](301)	
	B	PA-I[VRS](211)					
	C	PIC [SD](209)					

SUBJECTS	
Abbr.	Full name
PIC	Pharmaceutical Inorganic chemistry
PP-I	Pharmaceutics-I
PA-I	Pharmaceutical Analysis
HAP-I	Human Anatomy and Physiology
CS	Communication Skill
RM	Remedial Mathematics/Biology

FACULTY	
Abbr.	Full name
SD	Sachin Dhawale
SP	Snehal Pawar
VRS	Vishakha Shelke
PVB	Pallavi Bhosle
KBJ	Kishan Jaiswal
SL	Snehal Lande

CLASS INCHARGE
Pallavi Bhosle



MASTER TIME TABLE INCHARGE
Momin Aaliya Firdous

PRINCIPAL
Shreeyash Institute Of Pharmaceutical Education And Research, Aurangabad

DEPARTMENT OF PHARMACY
CLASS TIME TABLE (DIVISION-B)

Class: B. Pharmacy First year (First Semester)

Academic Year 2021-2022

w.e.f.: 5 / 1 / 2022

TIME	(MORNING) PRACTICAL			(AFTERNOON) THEORY			
	09:00am-1:00 pm	01:00 pm to 01:45 pm		01:45-02:45 pm	02:45-03:45 pm	03:45-04:45 pm	04:45-05:45 pm
DAY	Batch	Subject	R E C E S S				
MON	E	PIC [SMS](209)		PIC [SD] (302)	RM[SL] (301)	PIC [SD] Tutorial (302)	PCEUTICS-I [SP](302)
TUE	D E	PIC [SMS](209) CS[KBJ](203)		RM[SL] (301)	PCEUTICS-I [SP](302)	PIC [SD] (301)	PCEUTICS-I [SP] Tutorial (302)
WED	D E	CS[KBJ](203) PA-I[MYC](211)		PCEUTICS-I [SP](302)	PIC [SD] (302)	HAP-I[PVB] Tutorial(302)	PA-I[VRS] Tutorial (302)
THU	D	PA-I[MYC](211)		CONTACT/ MEETING HOUR			
FRI	D E	HAP-I[SK](312) PCEUTICS-I[MDK](013)		HAP-I [PVB] (302)	PA-I [VRS] (302)	HAP-I [PVB] (302)	CS[KBJ] (302)
SAT	D E	PCEUTICS-I[MDK](013) HAP-I[SK](312)		PA-I[VRS] (302)	HAP-I [PVB] (302)	PA-I [VRS] (302)	CS[KBJ] (301)

SUBJECTS	
Abbr.	Full name
PIC	Pharmaceutical Inorganic chemistry
PP-I	Pharmaceutical Analysis-I
PA	Pharmaceutical Analysis-II
HAP-I	Human Anatomy and Physiology-I
CS	Communication Skill
RM	Remedial Mathematics/Biology

FACULTY	
Abbr.	Full name
SD/SMS	Sachin Dhawale/ Sayyed Sir
SP/MDK	Snehal Pawar/Milind Kamble
VRS/MYC	Vishakha Shelke/ MinalChaudhari
PVB/SK	Pallavi Bhosle/ Sneha Khandale
KBJ	Kishan Jaiswal
SL	Snehal Lande

CLASS INCHARGE
Pallavi Bhosle

MASTER TIME TABLE INCHARGE
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DEPARTMENT OF PHARMACY

CLASS TIME TABLE (DIVISION-B)

Class: B. Pharmacy Second year (Third Semester)

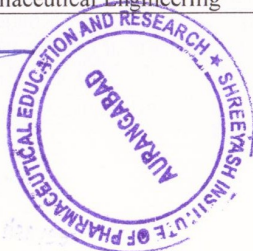
Academic Year 2021-2022

w.e.f.: 4/ 10/2021

TIME	(MORNING)THEORY				(AFTERNOON) PRACTICAL		
	9:00 -10:00am	10-11:15 am	11:15-12:15 pm	12:15-1:15 pm	01:15-2:00pm		2:00-6:00 pm
DAY					R E C E S S	Batch	Subject
MON	LIBRARY	CONTACT HOUR	PM(TJ) (302)	PE(PU) Tutorial(302)		D	PE(RK)[011]
TUE	ASSIGNMENT	PM(TJ) (302)	PE(PU) (302)	PM(TJ) Tutorial(302)		E	POC-II(AJ)[209]
WED	TG HOUR	PE(PU) (302)	PM(TJ) (302)	PE(PU) (302)		E	PP-I (VM)[013]
THU	LIBRARY	POC-II(MMK) (302)	PP-I (MYC) (302)	POC-II(MMK) Tutorial(302)		D	PM(SK)[310]
FRI	UNIT TEST	PP-I (MYC) (302)	POC-II(MMK) (302)	PP-I (MYC) Tutorial (302)		D E	POC-II(AJ)[209] PM(SK)[310]
SAT	ASSIGNMENT	POC-II(MMK) (302)	PP-I (MYC) (302)	UNIT TEST		D E	PP-I (VM)[013] PE(RK) [011]

SUBJECTS		FACULTY	
Abbr.	Full name	Abbr.	Full name
POC-II	Pharmaceutical Organic chemistry-II	MMK/AJ	Mahendra Khandare / Aditi Jotishi
PP-I	Physical Pharmaceutics-I	MYC/VM	Minal Chaudhari / Vinayak Mhaismale
PM	Pharmaceutical Microbiology	TJ/SK	Tushar Joshi/Sneha Khandale
PE	Pharmaceutical Engineering	PU/RK	Priti Undre / Rupesh Khalse

CLASS INCHARGE
Mahendra Khandare



MASTER TIME TABLE INCHARGE
Arundhati Deokar

PRINCIPAL

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**DEPARTMENT OF PHARMACY
CLASS TIME TABLE (DIVISION-A)**

Class: B. Pharmacy Second year (Third Semester)

Academic Year 2021-2022

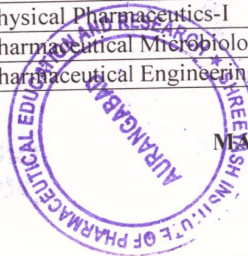
w.e.f:4 / 10 /2021

TIME	(MORNING)THEORY				(AFTERNOON) PRACTICAL		
	9:00 -10:00am	10-11:15 am	11:15-12:15 pm	12:15-1:15 pm	01:15-2:00pm		2:00-6:00 pm
DAY					R E C E S S	Batch	Subject
MON	LIBRARY	CONTACT HOUR	PP-I (MYC) (301)	POC-II (MMK)(301)		A B	POC-II(MMK)[209] PP-I (MYC)[013]
TUE	ASSIGNMENT	PP-I (MYC) (301)	POC-II(MMK) (301)	PP-I(MYC) Tutorial(301)		A B C	PP-I (MYC)[013] PM(TJ)[310] PE(PU)[011]
WED	TG HOUR	POC-II(MMK) (301)	PP-I (MYC) (301)	POC-II(MMK) Tutorial (301)		A B C	PM(TJ)[310] PE(PU)[011] POC-II(MMK)[209]
THU	LIBRARY	PE(PU) (301)	PM(TJ) (301)	PE Tutorial (PU)(301)		A B C	PE(PU) [011] POC-II(MMK)[209] PP-I (MYC)[013]
FRI	UNIT TEST	PM(TJ) (301)	PE(PU) (301)	PM(TJ) Tutorial(301)			
SAT	ASSIGNMENT	PE(PU) (301)	PM(TJ) (301)	UNIT TEST		C	PM(TJ)[310]

SUBJECTS	
Abbr.	Full name
POC-II	Pharmaceutical organic chemistry-II
PP-I	Physical Pharmaceutics-I
PM	Pharmaceutical Microbiology
PE	Pharmaceutical Engineering

FACULTY	
Abbr.	Full name
MMK	Mahendra Khandare
MYC	Minal Chaudhari
TJ	Tushar Joshi
PU	Priti Undre

mmk
CLASS INCHARGE
Mahendra Khandare



Arundhati
MASTER TIME TABLE INCHARGE
Arundhati Deokar

[Signature]
PRINCIPAL
Shreeyash Institute Of Pharmaceutical Education And Research, Aurangabad



DEPARTMENT OF PHARMACY
CLASS TIME TABLE (DIVISION-B)

Class: B. Pharmacy Third Year (Fifth Semester)

Academic Year: 2020-21

w.e.f.:17 / 8 /2021

			(MORNING) PRACTICAL		(AFTERNOON) THEORY			
TIME	08:00am-9:00 am		09:00am-1:00 pm	01:00 pm to 01:45 pm	01:45-02:45	02:45-03:45	03:45-04:45	04:45-05:45
DAY	Tutorial	Batch	Subject	R E C E S S				
MON	PCOG-II Tutorial [AVD](303)		CONTACT HOUR		PCOG-II [AVD](303)	P'COLOGY-III [PVB](303)	PJ[SD] (303)	MICRO[SD] (303)
TUE	PJ[SD] Tutorial (303)	D E	POC-II[SD](011) IP-I[SD](013)		P'COLOGY-III [PVB](303)	PCOG-II [AVD](303)	PJ[SD] (303)	MICRO[SD] (303)
WED	POC-II Tutorial [SD] (303)	D E	MICRO[HPK](310) POC-II[SD](011)		P'COLOGY-III [PVB](303)	PCOG-II [AVD](303)	P'COLOGY-III(303) Tutorial[PVB]	MICRO[SD] (303)
THU	MICRO Tutorial [SD] (303)	D E	P'COG[VRS](310) P'COLOGY-II[RBK](312)		M.CHEM[VRS] (303)	IP-I [SVN](303)	M.CHEM[VRS] Tutorial (303)	POC-II[SD] (303)
FRI		D E	P'COLOGY-II[RBK](312) P'COG[VRS](310)		IP-I [SVN](303)	PJ[SD] (303)	IP-I[SVN] Tutorial (303)	POC-II[SD] (303)
SAT		D E	IP-I[SD](013) MICRO[HPK](310)		M.CHEM[VRS](303)	IP-I [SVN](303)	M.CHEM[VRS] (303)	POC-II[SD] (303)

SUBJECTS	
Abbr.	Full Name
IP-I	Industrial Pharmacy-I
P'COG-II	Pharmacognosy-II
P'COLOGY-II	Pharmacology-II
M.CHEM-II	Medicinal Chemistry-II
PJ	Pharmaceutical Jurisprudance
POC-II	Pharmaceutical organic chemistry-II
MICRO	Microbiology

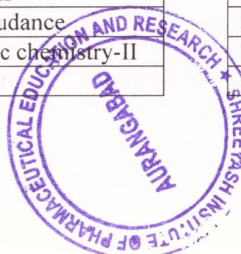
CLASS INCHARGE
Arundhati Deokar

FACULTY	
Abbr.	Full Name
SVN/ SD	Shraddha Nalawade/ Shubhangi Deshmukh
AVD/VRS	Arundhati Deokar/Vishakha Shelke
PVB/RBK	PallaviBhosle / Rupesh Khalse
VRS	Vishakha Shelke
SD	Sachin Dhawale
SD	Shubhangi Deshmukh
SD/HPK	Shubhangi Deshmukh /Hujeb Pathan

MASTER TIME TABLE INCHARGE
Arundhati Deokar

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DEPARTMENT OF PHARMACY
CLASS TIME TABLE (DIVISION-A)

Class: B. Pharmacy Third Year (Fifth Semester)

Academic Year: 2020-21

w.e.f.: 17 / 8 / 2021

TIME	(MORNING) PRACTICAL			(AFTERNOON) THEORY			
	09:00am-1:00 pm	01:00 pm to 01:45 pm		01:45-02:45	02:45-03:45	03:45-04:45	04:45-05:45
DAY	Batch	Subject	R E C E S S				
MON		CONTACT HOUR		M.CHEM-II [VRS] (314)	IP-I [SVN](314)	M.CHEM-II [VRS] Tutorial(314)	LIBRARY
TUE	A B	P*COLOGY-II[PVB](312) P*COG[AVD](310)		IP-I [SVN](314)	M.CHEM-II [VRS] (314)	IP-I[SVN] Tutorial(314)	UNIT TEST
WED	A B C	IP-I[SVN](013) P*COLOGY-II[PVB](312) P*COG[AVD](310)		PJ[SD] (314)	IP-I [SVN](314)	M.CHEM-II [VRS] (314)	PJ[SD] Tutorial(314)
THU	B	IP-I[SVN](013)		PCOG-II [AVD](314)	P*COLOGY-II [PVB](314)	PJ[SD] (314)	PCOG-II [AVD] Tutorial(314)
FRI	C	IP-I[SVN](013)		P*COLOGY-II [PVB](314)	PCOG-II [AVD](314)	P*COLOGY-II[PVB] Tutorial(314)	ASSIGNMENT
SAT	A C	P*COG[AVD](310) P*COLOGY-II[PVB](312)		PJ[SD] (314)	P*COLOGY-II [PVB](314)	PCOG-II [AVD](314)	UNIT TEST

SUBJECTS	
Abbr.	Full Name
IP-I	Industrial Pharmacy-I
P*COG-II	Pharmacognosy-II
P*COLOGY-II	Pharmacology-II
M.CHEM-II	Medicinal Chemistry-II
PJ	Pharmaceutical Jurisprudence

FACULTY	
Abbr.	Full Name
SVN/RK	Shraddha Nalawade
AVD	Arundhati Deokar
PVB	Pallavi Bhosle
VRS	Vishakha Shelke
SD	Sachin Dhawale

CLASS INCHARGE
Ms.Arundhati Deokar



MASTER TIME TABLE INCHARGE
Ms.Arundhati Deoka

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SHREEYASH PRATISHTHAN'S

FORMAT NO. STC/PR08/F/03

Shreeyash Technical Campus**SHREEYASH INSTITUTE OF PHARMACEUTICAL EDUCATION & RESEARCH****DEPARTMENT OF PHARMACY
CLASS TIME TABLE (DIVISION-A)**

Class: B. Pharmacy Final year (Seventh Semester)

Academic Year 2020-2021

w.e.f.: 17/ 8/2021

TIME	(MORNING)THEORY				(AFTERNOON) PRACTICAL		
	9:00 -10:00am	10-11:15 am	11:15-12:15 pm	12:15-1:15 pm	01:15-2:00pm	2:00-6:00 pm	
DAY					R	Batch	Subject
MON	LIBRARY	CONTACT HOUR	PP(SK) (303)	NDDS(VM) Tutorial(303)	R E C E S S	A	IMA(MS)[211]
TUE	ASSIGNMENT	PP(SK) (303)	NDDS(VM) (303)	PP(SK) Tutorial (303)		B	IMA(MS)[211]
WED	TG HOUR	NDDS(VM) (303)	PP(SK) (303)	NDDS(VM) (303)		C	IMA(MS)[211]
THU	LIBRARY	IMA(MS) (303)	IP-II (MDK) (303)	IMA(MS) Tutorial(303)			
FRI	UNIT TEST	IP-II (MDK) (303)	IMA(MS) (303)	IP-II (MDK) Tutorial(303)			
SAT	ASSIGNMENT	IP-II (MDK) (303)	UNIT TEST	IMA(MS) (303)			

SUBJECTS	
Abbr.	Full name
NDDS	Novel Drug Delivery system
PP	Pharmacy Practice
IMA	Instrumental Methods Of Analysis
IP	Industrial Pharmacy-II

FACULTY	
Abbr.	Full name
VM	Vinayak Mhaismale
SK	Sneha Khandale
MS	Mohamed Sayyed/ Milind Kamble
MDK	Milind Kamble

CLASS INCHARGE
Mohamed Sayyed

MASTER TIME TABLE INCHARGE
Arundhati Deokar

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**DEPARTMENT OF PHARMACY
CLASS TIME TABLE (DIVISION-B)**

Class: B. Pharmacy Final year (Seventh Semester)

Academic Year 2020-2021

w.e.f.: 17 / 8 /2021

TIME	(MORNING)THEORY				(AFTERNOON) PRACTICAL		
	9:00 -10:00am	10-11:15 am	11:15-12:15 pm	12:15-1:15 pm	01:15-2:00pm	Batch	2:00-6:00 pm Subject
DAY					R E C E S S		
MON	LIBRARY	CONTACT HOUR	IP-II(MDK) (314)	IMA(MS) Tutorial(314)			
TUE	ASSIGNMENT	IP-II(MDK) (314)	IMA(MS) (314)	IP-II (MDK) Tutorial(314)			
WED	TG HOUR	IMA(MS) (314)	IP-II(MDK) (315)	IMA(MS) (314)			
THU	LIBRARY	NDDS(VM) (314)	PP(SK) (314)	NDDS(VM) Tutorial(314)			
FRI	UNIT TEST	PP(SK) (314)	NDDS(VM) (314)	PP(SK) Tutorial(314)		D	IMA(MDK)[211]
SAT	ASSIGNMENT	NDDS(VM) (314)	UNIT TEST	PP(SK) (314)		E	IMA(MDK)[211]

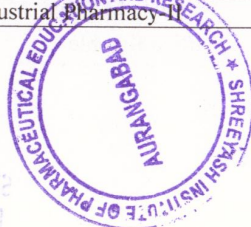
SUBJECTS	
Abbr.	Full name
NDDS	Novel Drug Delivery system
PP	Pharmacy Practice
IMA	Instrumental Methods Of Analysis
IP	Industrial Pharmacy-I

FACULTY	
Abbr.	Full name
VM	Vinayak Mhaismale
SK	Sneha Khandale
MS	Mohamed Sayyed/ Milind Kamble
MDK	Milind Kamble

CLASSINCHARGE
Mohamed Sayyed

MASTER TIME TABLE INCHARGE
Arundhati Deokar

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Ref. No. SYIPER/Admin /

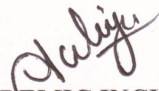
Date- 26-5-22


NOTICE

All the Teaching & Non-Teaching staff members and students are here by informed that our second semester academic will be start from 1st of June, 2022. All First year students be present and attend the class regularly.

All students should take note of this.




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Ref. No. SYIPER/Admin /

Date- 14-04-22

NOTICE

All the Teaching & Non-Teaching staff members and students are here by informed that our fourth semester academic will be start from 18th of April, 2022. All Second year students be present and attend the class regularly.

All students should take note of this.

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Ref. No. SYIPER/Admin /

Date-08-03-2022

NOTICE

All the Teaching & Non-Teaching staff members and students are here by informed that our six and eight semester academic will be start from 10th of March, 2022. All Third and Final year students be present and attend the class regularly.

All students should take note of this.


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Date - 10/12/2021


NOTICE

All the Teaching & Non-Teaching staff members and students are here by informed that our first semester academic will be start from 13th of Dec, 2021. All First year students be present and attend the class regularly.

All students should take note of this.


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Date - 25/9/2021


NOTICE

All the Teaching & Non-Teaching staff members and students are here by informed that our Third semester academic will be start from 27th of Sep, 2021. All ~~Second~~^{Third} year students be present and attend the class regularly.

All students should take note of this.


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NOTICE

Date - 1/10/2021

All the Teaching & Non-Teaching staff members and students are here by informed that our Third semester academic will be start from 4th of Oct, 2021. All First year students be present and attend the class regularly.

All students should take note of this.


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NOTICE


Date - 16/8/2021

All the Teaching & Non-Teaching staff members and students are here by informed that our Fifth semester academic will be start from 17th of August, 2020. All Second year students be present and attend the class regularly.

All students should take note of this.


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Affiliated to Dr. Babasaheb Ambedkar Technological University, Lonere & MSBTE Mumbai.

Date - 14/8/2021


NOTICE

All the Teaching & Non-Teaching staff members and students are here by informed that our Seventh semester academic will be start from 17th of August, 2020. All Second year students be present and attend the class regularly.

All students should take note of this.


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Syllabus for the Bachelor of Pharmacy (B. Pharm) Course

https://www.pci.nic.in/pdf/Syllabus_B_Pharm.pdf



TEACHING PLAN (THEORY)

Subject: IP-IClass: IIIrd yr Division: AUnit with Name : 1. PreformulationAllocated Hours: 07

Predicted Date of Commencement:

Predicted Date of Completion:

Sr. No.	Sub-Topic	Actual Dates of		Teaching Mode	Total Lect.	% cumulative syllabus completed
		of	Date			
1	Introduction to Preformulation, Goal and objectives	Commencement	17/8/21	Board & marker	01	2.17
		Completion	17/8/21			
2	Physical form, Particle size, shape and flow properties	Commencement	18/8/21	Board & marker	01	4.34
		Completion	18/8/21			
3	Solubility Profile	Commencement	23/8/21	Board & marker	01	6.52
		Completion	23/8/21			
4	Polymorphism	Commencement	24/8/21	Board & marker	01	8.69
		Completion	24/8/21			
5	Hydrolysis, oxidation, Reduction, Racemisation and Polymerization	Commencement	25/8/21	Board & marker	01	10.86
		Completion	25/8/21			
6	BCS classification of drug & its significant	Commencement	30/8/21	Board & marker	01	13.04
		Completion	30/8/21			
7	Application of Preformulation	Commencement	31/8/21	Board & marker	01	15.21
		Completion	31/8/21			
8		Commencement				
		Completion				
9		Commencement				
		Completion				
10		Commencement				
		Completion				


 Subject I/c


 Principal





TEACHING PLAN (THEORY)

Subject: IP-I Class: IIIrd yr Division: A

Unit with Name : 2. Tablets and Liquid orals Allocated Hours: 10 hrs.

Predicted Date of Commencement: _____ Predicted Date of Completion: _____

Sr. No.	Sub-Topic	Actual Dates of	Date	Teaching Mode	Total Lect.	% cumulative syllabus completed
1	Introduction Ideal characteristics & classification of tablet	Commencement	1/9/21	Board & marked	01	17.39
		Completion	1/9/21			
2	Excipients & formula of tablet	Commencement	6/9/21	Board & marked	01	19.56
		Completion	6/9/21			
3	Granulation mtd, Compression & processing Problem	Commencement	7/9/21	Board & marked	01	21.73
		Completion	7/9/21			
4	Equipments & tablet tooling, types of coating	Commencement	8/9/21	Board & marked	01	23.91
		Completion	8/9/21			
5	Formula & mtd. of coating, Equipment employed	Commencement	14/9/21	Board & marked	01	26.08
		Completion	14/9/21			
6	Defects in coating, QC test.	Commencement	15/9/21	Board & marked	01	28.26
		Completion	15/9/21			
7	Formulation & mfg of syrups & elixirs	Commencement	20/9/21	Board & marked	01	30.93
		Completion	20/9/21			
8	Formula & mfg of suspension & emulsion	Commencement	21/9/21	Board & marked	01	32.60
		Completion	21/9/21			
9	Filling & packaging	Commencement	22/9/21	Board & marked	01	34.78
		Completion	22/9/21			
10	Evaluation of liquid orals	Commencement	27/9/21	Board & marked	01	36.95
		Completion	27/9/21			


Subject I/c


Principal





TEACHING PLAN (THEORY)

Subject: <u>IP-I</u>		Class: <u>IIIrd yr</u>		Division: <u>A</u>		
Unit with Name : <u>3. Capsules and pellets</u>			Allocated Hours: <u>09 hrs</u>			
Predicted Date of Commencement:			Predicted Date of Completion:			
Sr. No.	Sub-Topic	Actual Dates of	Date	Teaching Mode	Total Lect.	% cumulative syllabus completed
1	Introduction, Prod'n of H ₂ C shell & size of capsule	Commencement	28/9/21	Board & marker	01	39.13
		Completion	28/9/21			
2	Filling, finishing & formulation of H ₂ C mfg defect	Commencement	29/9/21	Board & marker	01	41.30
		Completion	29/9/21			
3	Nature of shell and capsule content size of H ₂ C	Commencement	11/10/21	Board & marker	01	43.47
		Completion	4/10/21			
4	Imp of base adsorption & minimum gm factor	Commencement	12/10/21	Board & marker	01	45.65
		Completion	5/10/21			
5	Packaging, storage and stability testing of H ₂ C	Commencement	13/10/21	Board & marker	01	47.82
		Completion	6/10/21			
6	Quality control test for capsules	Commencement	18/10/21	Board & marker	01	50
		Completion	18/10/21			
7	Introduction of pellets	Commencement	19/10/21	Board & marker	01	52.17
		Completion	19/10/21			
8	Formulation requirement & pelletization process	Commencement	20/10/21	Board & marker	01	54.34
		Completion	20/10/21			
9	Equipments for manufacture of pellets	Commencement	25/10/21	Board & marker	01	56.52
		Completion	25/10/21			
10		Commencement				
		Completion				

Subject I/c

Principal





Format No. STC/PR09/F/6

TEACHING PLAN (THEORY)


Subject: IP-I Class: IIIrd yr Division: A

Unit with Name: G. Parenteral Product and ophthalmic preparations Allocated Hours: 10 hrs.

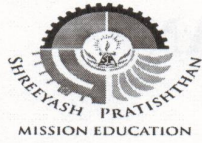
Predicted Date of Commencement: _____ Predicted Date of Completion: _____

Sr. No.	Sub-Topic	Actual Dates of	Date	Teaching Mode	Total Lect.	% cumulative syllabus completed
1	Defn types, adv and limitation of parenteral	Commencement	26/10/21		01	58.69
		Completion	26/10/21			
2	Preformulation factors, essential requirements, vehicle & additives	Commencement	27/10/21		01	60.86
		Completion	27/10/21			
3	Imp. of Isotonicity Product Procedure	Commencement	9/11/21		01	63.09
		Completion	9/11/21			
4	Production facilities and control, aseptic processing	Commencement	10/11/21	Board	01	65.21
		Completion	10/11/21			
5	Formulation of inj & sterile, powder, LVP & lyophilized product	Commencement	15/11/21	& market	01	67.39
		Completion	15/11/21			
6	Containers & closure selection, Filling & sealing of ampoules	Commencement	16/11/21		01	69.56
		Completion	16/11/21			
7	Vials & Infusion this QC test of parenteral product	Commencement	17/11/21		01	71.73
		Completion	22/11/21			
8	Formulation of eye drops, eye ointment & eye lotion	Commencement	22/11/21		01	73.91
		Completion	23/11/21			
9	Mfg. of Preparation, labelling & containers	Commencement	23/11/21		01	76.08
		Completion	24/11/21			
10	Evaluation of ophthalmic preparation	Commencement	24/11/21		01	78.26
		Completion	29/11/21			


Subject I/c


Principal





TEACHING PLAN (THEORY)

Subject: IP-IClass: IIIrd yrDivision: AUnit with Name: 5. Cosmetics, p^ractical
Aerosols and PackagingAllocated Hours: 10 hrs

Predicted Date of Commencement:

Predicted Date of Completion:

Sr. No.	Sub-Topic	Actual Dates of	Date	Teaching Mode	Total Lect.	% cumulative syllabus completed
1	Formula of lipsticks, shampoos, Cold cream and Vanishing cream	Commencement	29/11/21		01	80.93
		Completion	30/11/21			
2	Formula of Tooth Pastes, hair dyes & sunscreen	Commencement	30/11/21		01	82.60
		Completion	1/12/21			
3	Definition, Propellants containers & valves of aerosols	Commencement	1/12/21		01	84.78
		Completion	6/12/21			
4	Types of aerosol system, formula & mfg of aerosols	Commencement	13/12/21	Board	01	86.95
		Completion	7/12/21			
5	Evaluation, QC & stability studies of aerosols	Commencement	14/12/21	& marked	01	89.13
		Completion	8/12/21			
6	materials used for packaging of p ^r actical product	Commencement	15/12/21		01	91.30
		Completion	20/12/21			
7	Factor influencing choice of containers	Commencement	20/12/21		01	93.47
		Completion	21/12/21			
8	Legal and official requirement for containers	Commencement	21/12/21		01	95.65
		Completion	21/12/21			
9	stability aspects of packaging materials	Commencement	21/12/21		01	97.82
		Completion	23/12/21			
10	QC test of packaging.	Commencement	28/12/21		01	100
		Completion	25/12/21			

Subject I/c

Principal





Format No. STC/PR09/F/6

TEACHING PLAN (THEORY)

Subject: IP-I

Class: IIIrd yr

Division: B


Unit with Name: 1. Preformulation


Allocated Hours: 07 hrs

Predicted Date of Commencement:

Predicted Date of Completion:

Sr. No.	Sub-Topic	Actual Dates of	Date	Teaching Mode	Total Lect.	% cumulative syllabus completed
1	Introduction to Preformulation goal and objective	Commencement	20/8/21		01	2.17
		Completion	20/8/21			
2	Physical form, particle size, shape & flow property	Commencement	21/8/21		01	4.34
		Completion	21/8/21			
3	Solubility profile	Commencement	26/8/21	Board	01	6.52
		Completion	26/8/21			
4	Polymorphism	Commencement	27/8/21	& marker	01	8.69
		Completion	27/8/21			
5	Hydrolysis, oxidn Redn, Racemization, polymerization	Commencement	28/8/21		01	10.86
		Completion	28/8/21			
6	BCS classification of drug & its significance	Commencement	2/9/21		01	13.04
		Completion	2/9/21			
7	Application of Preformulation	Commencement	3/9/21		01	15.21
		Completion	3/9/21			
8		Commencement				
		Completion				
9		Commencement				
		Completion				
10		Commencement				
		Completion				


Subject I/c


Principal





TEACHING PLAN (THEORY)

Subject: <u>IP-I</u>		Class: <u>IIIrd yr</u>		Division: <u>B</u>		
Unit with Name: <u>2. Tablets and liquid orals</u>			Allocated Hours: <u>10 hrs</u>			
Predicted Date of Commencement:			Predicted Date of Completion:			
Sr. No.	Sub-Topic	Actual Dates of	Date	Teaching Mode	Total Lect.	% cumulative syllabus completed
1	Introdn, Ideal characteristics & classification of tab.	Commencement	<u>4/9/21</u>		01	17.39
		Completion	<u>4/9/21</u>			
2	Excipients & formuln of tablet	Commencement	<u>9/9/21</u>		01	19.56
		Completion	<u>9/9/21</u>			
3	Granuln mtd Compression & Processing Problem	Commencement	<u>11/9/21</u>		01	21.73
		Completion	<u>11/9/21</u>			
4	Equipments & tablet tooling, Types of Coating.	Commencement	<u>16/9/21</u>		01	23.91
		Completion	<u>16/9/21</u>			
5	Formuln & mtd of Coating equipment employed	Commencement	<u>17/9/21</u>	Board &	01	26.08
		Completion	<u>17/9/21</u>			
6	Defect in coating, qc test	Commencement	<u>18/9/21</u>	marked	01	28.26
		Completion	<u>18/9/21</u>			
7	Formulation & mfg of syrup & elixir	Commencement	<u>23/9/21</u>		01	30.43
		Completion	<u>23/9/21</u>			
8	Formuln & mfg of suspension & emulsion	Commencement	<u>24/9/21</u>		01	32.60
		Completion	<u>24/9/21</u>			
9	Filling & packaging	Commencement	<u>25/9/21</u>		01	34.78
		Completion	<u>25/9/21</u>			
10	Evaluation of liquid orals	Commencement	<u>30/9/21</u>		01	36.95
		Completion	<u>30/9/21</u>			

Subject I/c

Principal





Format No. STC/PR09/F/6

TEACHING PLAN (THEORY)

Subject: IP-I

Class: IIIrd yr

Division: B

Unit with Name : 3. Capsules & pellets

Allocated Hours: 09 hrs

Predicted Date of Commencement:

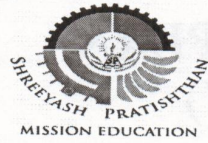
Predicted Date of Completion:

Sr. No.	Sub-Topic	Actual Dates of	Date	Teaching Mode	Total Lect.	% cumulative syllabus completed
1	Intro, prod ⁿ of H ₂ C shell & size of capsule	Commencement	11/10/21		01	39.13
		Completion	11/10/21			
2	Filling finishing & formula ⁿ of H ₂ C. mfg defect	Commencement	21/10/21		01	41.30
		Completion	7/11/21			
3	Nature of shell, capsule content & size of H ₂ C.	Commencement	14/11/21		01	43.47
		Completion	8/11/21			
4	Imp of base adsorption & minim/gm factor	Commencement	16/11/21		01	45.65
		Completion	9/11/21	Board		
5	Packaging storage & stability of H ₂ C	Commencement	21/11/21		01	47.82
		Completion	21/11/21	& marker		
6	QC test for Capsules	Commencement	22/11/21		01	50
		Completion	22/11/21			
7	Introduction of pellets.	Commencement	23/11/21		01	52.17
		Completion	23/11/21			
8	Formula ⁿ require-ment & pelletization process.	Commencement	28/11/21		01	54.39
		Completion	28/11/21			
9	Equipment for mfg of pellets.	Commencement	29/11/21		01	56.52
		Completion	29/11/21			
10		Commencement				
		Completion				

Subject I/c

Principal






TEACHING PLAN (THEORY)

Subject: IP-I Class: IIIrd yr Division: B

Unit with Name: G. Parenteral, pdts and ophthalmic preparations Allocated Hours: 10 hrs

Predicted Date of Commencement: Predicted Date of Completion:

Sr. No.	Sub-Topic	Actual Dates of	Date	Teaching Mode	Total Lect.	% cumulative syllabus completed
1	Defn, type, adv & limitation of Parenteral product	Commencement	30/10/21		01	58.69
		Completion	30/10/21			
2	Factors, essential requirements, vehicle & additives	Commencement	11/11/21		01	60.86
		Completion	11/11/21			
3	Imp of IS, tonicity, Production procedure	Commencement	12/11/21		01	63.04
		Completion	12/11/21			
4	Production facilities & control, aseptic processing	Commencement	13/11/21	Board	01	65.21
		Completion	13/11/21			
5	Formuln of inj & sterile powder, LVP & lyophilized pdt	Commencement	18/11/21	& marked	01	67.39
		Completion	18/11/21			
6	Containers & closure selection, filling & sealing of ampoule	Commencement	19/11/21		01	69.56
		Completion	19/11/21			
7	Vials & Infusion flts, Pc test of parenteral product	Commencement	20/11/21		01	71.73
		Completion	25/11/21			
8	Formuln of eye drop, eye ointment, eye lotion	Commencement	25/11/21		01	73.91
		Completion	26/11/21			
9	Mtd of Preparation, labelling & containers	Commencement	26/11/21		01	76.03
		Completion	27/11/21			
10	Evaluation of ophthalmic prepn	Commencement	27/11/21		01	78.26
		Completion	21/12/21			


Subject I/c


Principal





Format No. STC/PR09/F/6

TEACHING PLAN (THEORY)

Subject: IP-I

Class: IIIrd yr

Division: B

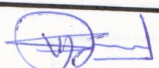
Unit with Name : 5. Cosmetics, pharmaceutical aerosol & packaging


Allocated Hours: 10 hrs.

Predicted Date of Commencement:

Predicted Date of Completion:

Sr. No.	Sub-Topic	Actual Dates of	Date	Teaching Mode	Total Lect.	% cumulative syllabus completed
1	Formulation of lipsticks, shampoos, cold cream & Vanishing cream	Commencement	2/12/21		01	80.43
		Completion	3/12/21			
2	Formulation of tooth paste, hair dyes & sunscreen	Commencement	3/12/21		01	82.60
		Completion	4/12/21			
3	Definition, Propellant, container & valve of aerosols	Commencement	4/12/21		01	84.78
		Completion	9/12/21			
4	Types of aerosols system, formulation of aerosols	Commencement	16/12/21	Board	01	86.95
		Completion	10/12/21			
5	Evaluation of GC & stability studies of aerosol	Commencement	17/12/21	& markers	01	89.13
		Completion	11/12/21			
6	Materials used for packaging of pharmaceutical pdt.	Commencement	18/12/21		01	91.30
		Completion	23/12/21			
7	Factor influencing choice of container	Commencement	23/12/21		01	93.47
		Completion	24/12/21			
8	Legal & official requirements for container	Commencement	24/12/21		01	95.65
		Completion	24/12/21			
9	Stability aspects of packaging material	Commencement	24/12/21		01	97.82
		Completion	25/12/21			
10	GC test of packaging	Commencement	25/12/21		01	100
		Completion	30/12/21			


Subject I/c


Principal





TEACHING PLAN (PRACTICAL)

Subject: IP-I

Class: IIIrd yr

Division: A

Sr. No.	Title	Date of Plan	Date of completion	Remark
11.	Preparation of creams. (Vanishing cream)	A-10/11/21	A-24/11/21	
		B-18/11/21	B-25/11/21	
		C-28/11/21	C-26/11/21	
12.	Evaluation of Glass Containers (as per IP)	A-17/11/21	A-1/12/21	
		B-25/11/21	B-2/12/21	
		C-26/11/21	C-3/12/21	

Principal





TEACHING PLAN (PRACTICAL)

Subject: IP-IClass: IIIrd yrDivision: A

Sr. No.	Title	Date of Plan	Date of completion	Remark
1.	Preformulation studies on Paracetamol / aspirin or any other drug	A-18/8/21	A-18/8/21	
		B-26/8/21	B-26/8/21	
		C-20/8/21	C-20/8/21	
2.	Preparation and evaluation of paracetamol tablet	A-25/8/21	A-25/8/21	
		B-2/9/21	B-2/9/21	
		C-27/8/21	C-27/8/21	
3.	Preparation & evaluation of aspirin tablet	A-1/9/21	A-1/9/21	
		B-9/9/21	B-9/9/21	
		C-3/9/21	C-3/9/21	
4.	Coating of tablets - film Coating of tablet granules	A-8/9/21	A-8/9/21	
		B-16/9/21	B-16/9/21	
		C-17/9/21	C-17/9/21	
5.	Preparation & evaluation of tetracycline capsules	A-15/9/21	A-15/9/21	
		B-23/9/21	B-23/9/21	
		C-24/9/21	C-24/9/21	
6.	Preparation of Calcium gluconate injection	A-22/9/21	A-22/9/21	
		B-30/9/21	B-30/9/21	
		C-29/9/21	C-29/9/21	
7.	Preparation of Ascorbic acid injection	A-1/10/21	A-1/10/21	
		B-14/10/21	B-21/10/21	
		C-22/10/21	C-22/10/21	
8.	QC test (as per IP) of marketed tablets and capsules	A-13/10/21	A-20/10/21	
		B-21/10/21	B-28/10/21	
		C-29/10/21	C-29/10/21	
9.	Preparation of eye drops / eye ointments	A-20/10/21	A-27/10/21	
		B-28/10/21	B-11/11/21	
		C-12/11/21	C-12/11/21	
10.	Preparation of Creams (Cold cream)	A-27/10/21	A-10/11/21	
		B-11/11/21	B-18/11/21	
		C-12/11/21	C-19/11/21	

Principal



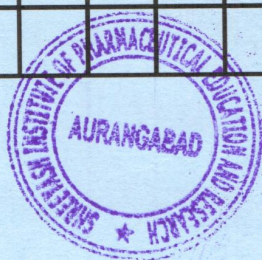
THEORY ATTENDANCE RECORD

SUBJECT:

CLASS:

DIV:

Roll No.	Date															
	Lec No.	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25
Total no. of students present:																



Subject I/c



Format No. STC/PR09/F/01

THEORY ATTENDANCE RECORD

SUBJECT: _____

CLASS: _____

DIV: _____

Roll No.	Date																
	Lec No.	26	27	28	29	30	31	32	33	34	35	36	37	38	39	40	41
Total no. of students present:																	



Subject I/c



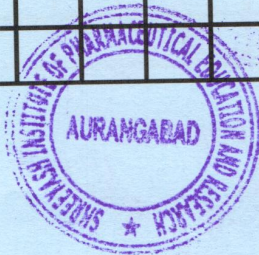
THEORY ATTENDANCE RECORD

SUBJECT:

CLASS:

DIV:

Roll No.	Date													Present attendance for semester	
		Lec No.	42	43	44	45	46	47	48	49	50	51	52	Ist	IInd
Total no. of students present:															



Subject I/c



SHREEYASH PRATISHTHAN'S
Shreeyash Technical Campus
 SHREEYASH INSTITUTE OF PHARMACEUTICAL EDUCATION & RESEARCH

FORMAT NO. STC/PR08/F/05



DEPARTMENT OF PHARMACY
LABORATORY OCCUPANCY REPORT

Academic Year 2021-2022 (EVEN SEM)

w.e.f.: 10/03/2022

Floor	Day	Monday		Tuesday		Wednesday		Thursday		Friday		Saturday		
G R O U N D	Lab/ Time	Morning (9:00 am - 1:00 pm)	Afternoon (2:00 pm - 6:00 pm)	Morning (9:00 am - 1:00 pm)	Afternoon (2:00 pm - 6:00 pm)	Morning (9:00 am - 1:00 pm)	Afternoon (2:00 pm - 6:00 pm)	Morning (9:00 am - 1:00 pm)	Afternoon (2:00 pm - 6:00 pm)	Morning (9:00 am - 1:00 pm)	Afternoon (2:00 pm - 6:00 pm)	Morning (9:00 am - 1:00 pm)	Afternoon (2:00 pm - 6:00 pm)	
	011	PROJECT WORK	PROJECT WORK	PROJECT WORK	PROJECT WORK	PROJECT WORK	PROJECT WORK	PROJECT WORK	PROJECT WORK	PROJECT WORK	PROJECT WORK	PROJECT WORK	PROJECT WORK	PROJECT WORK
	013	PROJECT WORK	PROJECT WORK	Batch-C [PP-II]	PROJECT WORK	Batch-E [PP-II]	PROJECT WORK	Batch-D [PP-II]	PROJECT WORK	PROJECT WORK	PROJECT WORK	Batch-B [PP-II]	PROJECT WORK	Batch-A [PP-II]
2nd	203	Batch-B [CA]	PROJECT WORK	Batch-E [CA]	PROJECT WORK	Batch-D [CA]	PROJECT WORK	Batch-C [CA]	PROJECT WORK	PROJECT WORK	PROJECT WORK	Batch-A [CA]	PROJECT WORK	
	209	PROJECT WORK	Batch-B [M.CHEM-III]	Batch-A [POC-I]	Batch-E [M.CHEM-III]	Batch-E [POC-I]	Batch-D [M.CHEM-III]	Batch-D [POC-I]	PROJECT WORK	Batch-C [POC-I]	Batch-A [M.CHEM-III]	Batch-B [POC-I]	Batch-C [M.CHEM-III]	
	211	Batch-E [Biochem]	Batch-B [M.CHEM-I]	Batch-D [Biochem]	Batch-C [M.CHEM-I]	Batch-B [Biochem]	Batch-A [M.CHEM-I]	PROJECT WORK	PROJECT WORK	Batch-A [Biochem]	Batch-D [M.CHEM-I]	Batch-C [Biochem]	Batch-E [M.CHEM-I]	
	213	PROJECT WORK	Batch-A [P.COLOGY-III]	PROJECT WORK	Batch-C [P.COLOGY-III]	PROJECT WORK	Batch-B [P.COLOGY-III]	PROJECT WORK	PROJECT WORK	PROJECT WORK	Batch-E [P.COLOGY-III]	PROJECT WORK	Batch-D [P.COLOGY-III]	
3rd	310	PROJECT WORK	Batch-A [P.COG-I]	PROJECT WORK	Batch-B [P.COG-I]	PROJECT WORK	Batch-C [P.COG-I]	PROJECT WORK	PROJECT WORK	PROJECT WORK	Batch-E [P.COG-I]	PROJECT WORK	Batch-D [P.COG-I]	
	312	Batch-A [HAP-II]	Batch-E [HDT]	Batch-B [HAP-II]	Batch-D [HDT]	Batch-C [HAP-II]	Batch-C [HDT]	PROJECT WORK	PROJECT WORK	Batch-D [HAP-II]	Batch-B [HDT]	Batch-E [HAP-II]	Batch-A [HDT]	
4th	414	PROJECT WORK	Batch-D [P.COLOGY-I]	PROJECT WORK	Batch-A [P.COLOGY-I]	PROJECT WORK	Batch-E [P.COLOGY-I]	PROJECT WORK	PROJECT WORK	PROJECT WORK	Batch-C [P.COLOGY-I]	PROJECT WORK	Batch-B [P.COLOGY-I]	

Valiya
 Academic In-Charge
 Shreeyash Institute of Pharmaceutical
 Education And Research
 Aurangabad



SPM
 PRINCIPAL
 Shreeyash Institute of Pharmaceutica
 Education and Research, Aurangabad



SHREEYASH PRATISHTAN'S
Shreeyash Technical Campus
 SHREEYASH INSTITUTE OF PHARMACEUTICAL EDUCATION & RESEARCH

FORMAT NO. STC/PR08/F/05




DEPARTMENT OF PHARMACY
LABORATORY OCCUPANCY REPORT


Academic Year 2021-2022 (ODD SEM)

w.e.f: 17/ 08/2021

Floor	Day	Monday		Tuesday		Wednesday		Thursday		Friday		Saturday	
		Lab/Time	Morning (9:00 am-1:00 pm)	Afternoon (2:00 pm - 6:00 pm)	Morning (9:00 am-1:00 pm)	Afternoon (2:00 pm - 6:00 pm)	Morning (9:00 am-1:00 pm)	Afternoon (2:00 pm - 6:00 pm)	Morning (9:00 am-1:00 pm)	Afternoon (2:00 pm - 6:00 pm)	Morning (9:00 am-1:00 pm)	Afternoon (2:00 pm - 6:00 pm)	Morning (9:00 am-1:00 pm)
G R O U N D	011	PROJECT WORK	Batch-D PE	Batch-E IP-I	Batch-C PE	Batch-E IP-I	Batch-B PE	Batch-E IP-I	Batch-A PE	Batch-C IP-I	PROJECT WORK	Batch-D IP-I	Batch-E PE
	013	Batch-B PH-I	Batch-B [PP-I]	Batch-C PH-I	Batch-A [PP-I]	PROJECT WORK	Batch-E [PP-I]	Batch-A PH-I	Batch-C [PP-I]	Batch-E PH-I	PROJECT WORK	Batch-D PH-I	Batch-D [PP-I]
	203	PROJECT WORK	PROJECT WORK	Batch-E [CSK]	PROJECT WORK	Batch-D [CSK]	PROJECT WORK	Batch-C [CA]	PROJECT WORK	Batch-B [CSK]	PROJECT WORK	Batch-A [CSK]	PROJECT WORK
2 nd	209	Batch-E PIC	Batch-A [POC-II]	Batch-D PIC	Batch-E [POC-II]	Batch-B PIC	Batch-C [POC-II]	PROJECT WORK	Batch-B [POC-II]	Batch-A PIC	Batch-D [POC-II]	Batch-C PIC	PROJECT WORK
	211	PROJECT WORK	Batch-A IMA	Batch-A PA-I	Batch-B IMA	Batch-E PA-I	Batch-C IMA	Batch-D PA-I	PROJECT WORK	Batch-C PA-I	Batch-D IMA	Batch-B PA-I	Batch-E IMA
	213	PROJECT WORK	PROJECT WORK	PROJECT WORK	PROJECT WORK	PROJECT WORK	PROJECT WORK	PROJECT WORK	PROJECT WORK	PROJECT WORK	PROJECT WORK	PROJECT WORK	PROJECT WORK
3 rd	310	PROJECT WORK	PROJECT WORK	Batch-B [P.COGE-II]	Batch-B PM	Batch-C [P.COGE-II]	Batch-A PM	Batch-D [P.COGE-II]	Batch-D PM	Batch-E [P.COGE-I]	Batch-E PM	Batch-A [P.COGE-II]	Batch-C PM
	312	Batch-A [HAP-I]	PROJECT WORK	Batch-B [HAP-I]	PROJECT WORK	Batch-C [HAP-I]	PROJECT WORK	PROJECT WORK	PROJECT WORK	Batch-D [HAP-I]	PROJECT WORK	Batch-E [HAP-I]	PROJECT WORK
4 th	414	PROJECT WORK	PROJECT WORK	Batch-A [P.COLOGY-II]	PROJECT WORK	Batch-B [P.COLOGY-II]	PROJECT WORK	Batch-E [P.COLOGY-II]	PROJECT WORK	Batch-D [P.COLOGY-II]	PROJECT WORK	Batch-C [P.COLOGY-II]	PROJECT WORK


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 Aurangabad




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SHREEYASH PRATISHTHAN'S
Shreeyash Technical Campus

FORM T NO. STC/PR08/F/05



SHREEYASH INSTITUTE OF PHARMACEUTICAL EDUCATION & RESEARCH

LAB TIME TABLE

LAB NAME: Pharmaceutical Analysis

ACADEMIC YEAR: 2021-2022

HALL NO: 211

W. E. F. :-.....17/8/2021...../ Issue:

LAB INCHARGE : Mr.Mohammad Sayeed/Mr. Milind Kamble/ Ms. Vishakha Shelke/ Minal Chaudhari

SEMESTER: I & VII

DAY / TIME	9:00 am-1:00 pm	01:00 am- 1.45 pm	2:00pm-6:00 pm
MON	←-----→		IMA (Batch A) (MS)
TUE	PA-I (Batch A) (VRS)	RECESS	IMA (Batch B) (MS)
WED	PA-I (Batch E) (MYC)		IMA (Batch C) (MS)
THU	PA-I (Batch D) (MYC)		Project Work
FRI	PA-I (Batch C) (VRS)		IMA (Batch D) (MDK)
SAT	PA-I (Batch B) (VRS)		IMA (Batch E) (MDK)

Faculty Initials	Faculty Full Name	Sign
MS	Mr Mohammad Sayeed	
MDK	Mr. Milind Kamble	
VRS	Ms. Vishakha Shelke	
MYC	Ms. Minal Chaudhari	

HOD



PRINCIPAL
Shreeyash Institute of Pharmaceutical
Education and Research, Aurangabad.

Shreeyash Institute of Pharmaceutical Education & Research

AURANGABAD

Certificate

This is to certify that

Student Name: _____

(Seat No :) _____ has carried out the required practical's work prescribed by the University of *Dr. Babasaheb Ambedkar Technological University, Lonere for Second year B. Pharmacy course* in the subject _____ during academic year _____ & this journal represents his/her work within four walls of this Institute.

Date:

**Subject
In- charge**

**External
Examiner**

Principal

Introduction to Qualitative Analysis

Introduction

Qualitative Organic Analysis is an essential part for the systematic identification of an unknown organic compound. It includes

- 1) Preliminary test
- 2) Detection of elements
- 3) Functional group detection
- 4) Determination of physical constant

In preliminary tests – Includes Physical characteristics, solubility tests, Ignition Test and reaction with KMnO_4 , Br_2 , Neutral FeCl_3 are carried out on the basis of the preliminary tests we can conclude whether the organic compound is saturated or unsaturated.

In elemental analysis – Identification of the organic compound containing the elements nitrogen, sulphur, halogens, Oxygen such as chlorine, bromine or iodine

In Functional group analysis – On the basis of the elements found, groups are classified to carry out tests for detection of functional group. In this detection we come to know the given organic compound from the determination of the physical constant (boiling point/ melting point) we can confirm the given compound. On the basis of elements present in the functional group there are three types of functional groups.

Group 1: Compounds which contains **C,H,O**

- a) Acid: Carboxylic Acid
- b) Phenols: Naphthol, Cresol
- c) Neutrals: Alcohols, Aldehydes, Ketones, Esters, Ethers, Carbohydrates, Acid Anhydrides, Quinones, Hydrocarbons.

Group 2: Compounds which contains **C,H,O** and **N**

- a) Acid: Amino, Nitro Acid
- b) Phenols: Nitro Phenols
- c) Base: Amines.
- d) Neutrals: Amides, Imides, Nitriles, Anilides, Nitrohydrocarbons

Group 3: Compounds which contains **C,H,O,N** and **S**

- a) Acid: Sulphonic Acid
- b) Neutral: Thiourea, Sulphonamids, Thiols

Introduction to Qualitative Analysis

Group 4: Compounds which contains **C,H,O**, and **Halogens**

- a) Acid: Halo Acid
- b) Neutral: Halo Hydrocarbons, Halo-aldehydes, Haloketones

In physical constant analysis: Melting point is determined as confirmatory test for that particular compound.

1. Preliminary Examination :

a) Physical characteristics

Knowledge of the physical state of organic compound is often a key factor (eg. Many common esters re liquid) n aid to choice or purification method is given by the state of the compound, wherein liquids are usually purified by distillation and solids are purified by recrystallization

1) **State:** Note whether the unknown substance is liquid or solid. If solid then whether it is crystalline or amorphous.

2) **Colour:** Many liquids and solids are definitely colored because of the presence of chromophoric groups in the molecule. Many nitro compounds, quinine, azo compounds, stable carbocations, carbanion and compounds with extended conjugated system are coloured. The colour of some compounds is due to impurities, frequently these are product by slow oxidation of the compounds by oxygen in the air. Aniline, for example, on prolonged storage is usually reddish brown but freshly distilled sample is nearly colourless.

Color of compound

Test	Observation	Inference
Colour of compound	In case of Colorless Solid	Carbohydrate, Simple Phenol, aliphatic higher carboxylic acids, acids-salts, amides, polyhydric phenols
	In case of Colored Solid	
	If yellow to orange	Iodoform, Nitro compound, Quinones, Diketones.
	If Yellow to pink	Napthols, Napthylamines
	If Red	Azo compounds, Methyl orange, β -napthaquinone.
	If Greenish	P-nitroso Compounds, quinhydrones
	If Brown	Amino phenols, Higher aromatic amines
Test	Observation	Inference

Introduction to Qualitative Analysis

	In case of colorless liquid	Alcohols, aldehydes, ketones, simple esters, lower ethers, aromatic hydrocarbons, and their simple halogens derivatives, lower aliphatic acids and their anhydrides, simple amines
	In case of colored Liquids	
	If Reddish Brown	Amines
	If Yellowish to Orange	Nitro Compounds, Diketones

3) **Odour:** Many types of organic compounds have characteristics odour. It is not possible to describe odour in a precise manner, but the student become familiar with the odours of common compounds. Moreover the odour is more pronounced in the lower molecular members of class, because these are more volatile

Test	Observation	Inference
Odour of compound	In case of phenolic odour	Phenol, Naphthols, cresols, derivatives of salicylic acid, bromobenzene.
	In case of Pleasant odour	lower aliphatic Alcohols , esters, ethers, aromatic hydrocarbons, halogens derivatives of aromatic hydrocarbons
	If fishy smell	Amines
	If Rotten eggs/ garlic like smell	Heteroatom sulphur containing compounds.
	If disagreeable smell	Azo compounds, Methyl orange, β -naphthaquinone.
	If Greenish	P-nitroso Compounds, quinhydrones
	If Brown	Amino phenols, Higher aromatic amines
	In case of colorless liquid	Alcohols, aldehydes, ketones, simple esters, lower ethers, aromatic hydrocarbons, and their simple halogens derivatives, lower aliphatic acids and their anhydrides, simple amines
	In case of colored Liquids	
	If Reddish Brown	Amines
	If Yellowish to Orange	Nitro Compounds, Diketones

Introduction to Qualitative Analysis

b) Ignition test

1. Heating on porcelain piece

In **aromatic compound**, the proportion of carbon and hydrogen is almost same. **Aromatic compounds** have excess % of C in their molecule so they burn with **smoky flame**

In **Aliphatic compound** the hydrogen are more compared to carbon. **Aliphatic compounds** having less % of C burn with **non-smoky flame**

(Exception: The aromatic compound which has more % also gives non sooty flame Eg: sulphosalicylic acid)

2. Beilstein test :

Copper foil or wire is heated (**without substance**) on flame to oxidize it to copper oxide. Copper oxide then reacts with halogen to give copper halide (CuX) which is volatile in nature and further dissociates on heating to give copper metal. **Green flame** is due to burning of copper metal. Test confirms presence of **Halogens**.

Reaction

The Beilstein's test fails to detect fluorine since CuF is non volatile.

Some halogen free compounds also responds this test viz **urea, thiourea, amid some pyridines, quinolines acids, cyano compounds** etc because these halogen free compounds form cuprous cyanide which is volatile and decompose to copper, which burns with green flame.

c) Unsaturation test

Unsaturated compounds contains double or triple bond in their structure eg. Alkenes, alkynes, benzene and its derivatives.

Alkene gives the test for unsaturation and decolorizes the KMnO₄ solution as well as Br₂ solution because of their addition reaction with MnO₄ and Br₂ to form diols and dibromo products respectively.

Benzene does not decolorize KMnO₄ solution even though it is having three double bonds in its structure because it utilizes 36 cal / mol energy in resonance and became stable compound. So benzene as well as other aromatic compounds behaves as saturated compounds.

Benzene substituted with activating group (electron releasing group like – OH- NHCOCH₃, - NH₂, - OCH₃) **reacts readily with bromine & decolorizes it**. It is an electrophilic aromatic substitution reaction. During this reaction, HBr gas is evolved, presence of which can be easily detected by congo red paper. HBr being a water soluble acid can only be evolved in bromine in CCl₄ reagent.

Introduction to Qualitative Analysis

Benzene substituted with deactivating groups (electron withdrawing groups like $-\text{NO}_2$, $-\text{CHO}$, $\text{C}=\text{O}$, $-\text{X}$, $-\text{COOR}$, $-\text{COOH}$, $-\text{CN}$) **does not decolorizes the bromine solution.**

Compounds containing **easily oxidizable functional groups** like alcoholic $-\text{OH}$, phenolic $-\text{OH}$, $-\text{CHO}$, $-\text{CO}$) also **decolorizes the KMnO_4 solution**

The presence of the unsaturation can be established by two ways

1) Action of KMnO_4 solution

2) Action of Br_2 water

1) Action of KMnO_4 solution

Potassium permanganate solution is most important oxidizing agent in neutral, acidic as well as alkaline medium

a) Reaction of KMnO_4 with alkenes

Potassium permanganate solution acts on an alkene to produce glycol. On further addition of the reagent and heating of the reaction mixture, further oxidation and cleavage of the carbon chain takes place.

b) Reaction of KMnO_4 with phenols

Phenols reduce potassium permanganate solution and undergo oxidation to quinines. The brown colour obtain after the reaction is due to Mn^{+4} (MnO_2) while Mn^{+7} (KMnO_4) compound is purple colour

Eg. Aldehydes, Phenols, Amines, formic acid its esters, thiols, thoesters etc.

(Note: Decolourisation of KMnO_4 solution is either due to unsaturation or oxidation of easily oxidisable group)

2) Action of Br_2 water:

a) Reaction of Br_2 with alkenes

This is electrophilic addition to $\text{C}=\text{C}$ bond. The first step is transfer of bromine to the olefin and formation of bromonium ion

Introduction to Qualitative Analysis

The second step is the reaction of the bromonium ion with bromide ion (Br^-) to give dibromoalkane. Due to steric hindrance, bromide ion attacks to the opposite side of the plane of the double bond, which results in a trans configuration. This is trans addition of bromine to the double bond.

b) Reaction of Br_2 with phenols

Phenols (having $-\text{OH}$, an electron donating functional group) are more reactive towards electrophilic aromatic substitution than benzene. Phenol is easily brominated under mild conditions and discharges the colour of bromine solution. Bromonium ion (Br^+), an electrophile, attacks at the ortho as well as the para positions of phenols and forms tribromosubstituted phenols.

d) Solubility Test:

2. Element Detection Test

Principle

Organic compounds consist of two or more elements. Apart from carbon and hydrogen, nitrogen, sulphur, and halogens can also be present, which have to be detected in organic compounds. Organic compounds are non-ionic in nature. Therefore, in order to detect the element in an organic compound, it is necessary to convert them into water-soluble ionizable inorganic substances. So that the ionic test or inorganic qualitative analysis may be applied. This conversion may be accomplished by several methods, but the best procedure is to fuse the organic compound with metallic sodium. In this way sodium cyanide, sodium sulphide, and sodium halides are formed, which are readily identified. Thus organic compounds containing these elements when strongly heated with elemental sodium lead to the following:

General Reaction



Procedure

A dry, clean, small freshly cut piece of sodium metal is placed in a sodium fusion tube. Heat at the bottom of the sodium fusion tube gently so that the sodium metal gets melted. Then pour a small amount of organic compound (0.1 to 0.2 ml) in the sodium fusion tube. Further heat the tube gently and later on strongly at the bottom so that the tube becomes red hot. The red hot tube is immediately dipped into a porcelain dish containing 10-20 ml of warm distilled water. Perform the same procedure for 2-3 sodium fusion tubes. Crush all the sodium fusion tubes with warm water and boil for 10 min and filter. The extract is then filtered; it is also known as sodium fusion extract.

Introduction to Qualitative Analysis

It is essential to use an excess of sodium, otherwise if sulphur & nitrogen are both present NaSCN, sodium thiocyanate, may be produced. In the test for nitrogen it may give Red colouration with ferric ion, but no Prussian blue colouration, since there will be no free cyanide ions. With excess of sodium the thiocyanate, if formed.

The filtered alkaline solution, Resulting from the action of water upon sodium fusion is treated with ferrous sulphate & thus sodium ferrocyanide

Upon boiling, alkaline ferrous salt solution, some ferric ions are inevitably produced by the action of the air, upon addition of dilute sulphuric acid, thus dissolving the ferrous & ferric hydroxide the ferrocyanide reacts with ferric salt producing ferric ferrocyanide.

2) Carboxylic Acid :- (R/Ar-COOH)

i) **Sodium Bicarbonate Test** :- IN this test effervescence of CO_2 comes & then substance dissolves the reappearance of the substance by adding dilute HCl (till acidic)

ii) **Ester Test** :- In this test fruity smell of ester is occurs. When 0.2 gm compound+ 3ml ethyl alcohol & 2 drops of conc. H_2SO_4 are added. After heating it gives fruity smell of ester.

3) Ferric Chloride Test (Phenols):-

In this 0.2 ml of compound + 1ml water or 1 ml ethanol + few drops aqueous FeCl_3 solution.

It gives appearance of violet, green, pink or red color.

If violet colour appears then phenol is present.

4) Phthalein Test (Phenol) :-

0.2 gm of compound in dry test tube + 0.2gm phthalic anhydride + 2 drops of conc. H_2SO_4 warm it to fuse. Cool then add dilute 2ml NaOH solution dropwise till colour is produced. It gives green, or yellowish, red, blue colour. According to colour solution it indicates the which phenol is present.

5) Liebermann's Nitroso Test :-

0.2 gm of compound +1ml conc. HCl in dry test tube + few crystals of sodium nitrite (NaNO_2).

Dilute with 5ml water. Finally add 10% NaOH solution 3-5 ml & observe colour change.

It gives Red/ intense green/ blue colour/ red/ dirty green colour. Which indicates phenol is present or not.

6) Neutral Test Solution :-

Introduction to Qualitative Analysis

0.5 gm compound mix with 1ml water+ 1drop of phenol pthalein solution +0.2 ml of ammonia solution. Boil to remove excess of ammonia. Cool & add few drops of aqueous FeCl_3 solution.

It is distinguish test for Acids. Such as if it gives buff coloured ppt means benzoic or phthalic acid.

Yellow coloured ppt – cinnamic acid

Violet coloured ppt- salicylic acid

Faint reddish coloured ppt – acetyl salicylic acid

Deep yellow coloured ppt – succinic or acetic acid

No change in colour – oxalic acid

Red colour ppt – acetic acid

7) Lucas Test :-

This test is performed for distinguish between primary, secondary & tertiary alcohols. In this test 1ml compound + 5-7 ml lucas reagent shake well & note time required for separation of 2 distinct layers.

If layers separate immediately then tertiary alcohol present, if layers separate takes 5-10 minutes then secondary or clear homogenous solution obtained then primary alcohol present.

8) 2,4 Dinitrophenyl Hydrazine Test :-

0.2 gm compound + 2 ml of methanol + 0.5 of 2,4 DNP + 0.5 ml of conc. H_2SO_4

It gives yellow/ orange or red crystalline precipitate. It indicates that carbonyl group is present.

9) Schiff's Test :-

0.2 gm compound + 2 ml schiff's reagent & shake well. In this test pink colour develops on standing. Which indicates that aromatic aldehyde is present. This test is negative for Ketones & some aromatic aldehydes e.g :- Vanillin.

10) Tollen's Test :-

In this test 2 ml tollen's reagent in test tube + 0.2 ml along side of tube then heat on water bath for 5 min without disturbing the test tube.

A shining silver mirror is deposited at bottom of test tube or grey black ppt. indicates that aldehyde group is confirmed.

11) Benedict's Test :-

Under alkaline conditions, reduce cupric ions to, cuprous form this is the basis for Benedict's test.

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0.2 ml of compound + 2ml of Benedict's solution. boil for 2 min & cool. It gives red ppt which confirms that aliphatic aldehydes is present.

12) Hydroxamic Acid Test :-

0.2 ml compound + 10% NaOH solution (till alkaline) heat to boil. Cool the soln & neutralize by dilute HCl add 1-2 drops of aqueous FeCl_3 soln. gives red violet colour Which confirms that Ester group is present.

13) Iodine Test :-

In this test, 0.2 ml compound + 5 ml benzene + 5 ml of very dil. Soln of Iodine in benzene, shake well. In this test a brown tint appears. Indicates that Ether is present.

14) Molish Test :-

This is the general reaction for all carbohydrates. In this test 0.2 gm compound + 2 ml Molisch's reagent, shake well & add 1 ml conc. H_2SO_4 from the walls of the tube. Violet red colouration formed at the junction of the two layers. When treated with H_2SO_4 , carbohydrates undergoes dehydration to yield furfural & its derivatives, α - naphthol of molisch reagent condenses with these furfural derivatives to form coloured compound.

15) Ninhydrine Test :-

This test consist 0.1 gm compound + 2 ml water + 1 ml ninhydrin solution. It gives blue violet colour, then α or β amino acid is confirmed.

16) Azodye test :

In this test, take 3 test tubes, in the first test tube take 0.2 g compound +2 ml of dilute HCL. cool to 0-5 $^{\circ}$ c. in the second test tube take 0.2g of NaNO_3 +2 ml water, cool to 0 $^{\circ}$ c-5 $^{\circ}$ c. mix these two test tubes. to this add a well cooled solution of 0.2g of β -naphthol in 5ml of 10% NaOH dropwise and stir well. then there is formation of an orange-red dye which confirms that aromatic amino acid is present. e.g Anthranilic acid.

17) Hinsberg Test:

0.2g of compound + 2ml pyridine +2ml freshly prepared 2% aqueous NaOH solution. shake well and add 2 drops of benzene sulphonyl chloride. If it gives yellow colour after shaking then primary amine is confirmed. If orange colour then secondary amine confirmed. If purple colour then tertiary amine confirmed.

18) P-Nitrosamine test:

0.2 g compound +2ml dilute HCL and cool to 0 $^{\circ}$ c. Now add cold NaNO_2 solution dropwise with shaking. If orange colour is formed, then tertiary amine is confirmed.

19) Nitrous acid test :

Introduction to Qualitative Analysis

0.2g compound in test tube +2ml dilute HCL+2ml NaNO_2 solution,if no effervescence observed then boil off N_2 from the above test.If brisk effervescence due to evolution of N_2 Amides confirmed.If clear solution amides of aliphatic acid is confirmed.If white Ppt amides of aromatic acid is confirmed.

20) Alkaline hydrolysis:

In this test,0.2g/0.2 ml compound+5ml of 10%NaOH solution and boil it.Evolution of ammonia which can be tested with rod dipped in concentrated HCL,white fumes occur.Indicates that Amides,Imides or Nitrile group is present.

Experimental Qualitative Analysis

EXPERIMENT NO: 01

DATE:

AIM: To carry out Qualitative Analysis of the given organic compounds

REFERENCES:

- 1) A laboratory handbook of Organic Qualitative analysis and separations – V.S Kulkarni and S. P. Pathak
- 2) Experimental Pharmaceutical Organic Chemistry A bench top manual by K.S. Jain , P.B.Maniyar and T.S, chitre

REQUIREMENT:

CHEMICALS: Saturated sodium bicarbonate, 10 %NaOH 50 % HCl , concentrated H_2SO_4 ,

$KMnO_4$, bromine water, aqueous ferric chloride 10 % sodium carbonat , carbon tetrachloride,

Alcohol copper foil

APPARATUS:

Test tubes, test tube stand, test tube holder, tong

Sr No	TEST	OBSERVATION	INFERENCE
A	State		
	Nature		
	Colour		
	Odour		
B	Heating on copper foil		
C	Solubility		
	Solubility in water		
	Solubility in saturated $NaHCO_3$ solution		

Experimental Qualitative Analysis

	Solubility in 10% NaOH Solubility in 50% HCl		
D	Treatment with Conc H_2SO_4		
E	Saturation Test Action of $KMnO_4$ Action of Br_2 Water		
F	Neutral Ferric Chloride Test		

Result:

Experimental Qualitative Analysis

Element Detection Test

Sr No	TEST	OBSERVATION	INFERENCE
A	Test For Nitrogen		
B	Test For Sulphur Lead Acetate Test Sodium Nitroprusside Test		
C	Test For Nitrogen and Sulphur		
D	Test For Halogen		
E	Differentiation Between Halogens		

Experimental Qualitative Analysis

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Result:

Functional Group Test:

Sr No	TEST	OBSERVATION	INFERENCE
A			
B			
C			
D			
E			

Experimental Qualitative Analysis

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Experimental Qualitative Analysis

Sr No	TEST	OBSERVATION	INFERENCE
F			
G			
H			
I			
J			

Synthesis

Experimental Qualitative Analysis

RESULT

<u>Name</u>	
<u>IUPAC nomenclature</u>	
<u>Structure</u>	
<u>Elements Present</u>	
<u>Function Group Present</u>	
<u>Physical Constant</u>	


List of various teaching pedagogies

(2021-22)

1. **Lecture:** Traditional method where the teacher presents information to students orally.
2. **Demonstration:** Showing students how to do something, often with visual aids or practical examples.
3. **Discussion:** Encouraging students to participate in dialogue, share ideas, and debate concepts.
4. **Collaborative Learning:** Group-based activities where students work together to achieve learning goals.
5. **Project-Based Learning (PjBL):** Similar to PBL, but focused on completing a project to demonstrate learning.
6. **Flipped Classroom:** Students learn new content at home through videos or readings, and then do activities or discussions in class.
7. **Inquiry-Based Learning:** Students explore questions, problems, or scenarios to develop understanding.
8. **Experiential Learning:** Hands-on learning through direct experience and reflection.
9. **Constructivism:** Students construct their understanding and knowledge through experiences and reflecting on those experiences.
10. **Socratic Method:** A form of cooperative argumentative dialogue between individuals, based on asking and answering questions to stimulate critical thinking.
11. **Differentiated Instruction:** Adapting teaching methods and curriculum to individual student needs and learning styles.
12. **Direct Instruction:** Teacher-centered approach where the teacher transmits information directly to students, often using structured lesson plans.
13. **Holistic Education:** Focuses on the development of the whole person, including intellectual, emotional, social, physical, artistic, creative, and spiritual dimensions.
14. **Outdoor Education:** Learning experiences that take place in natural settings, emphasizing environmental education and experiential learning.
15. **Virtual Learning:** Education delivered electronically through various forms of multimedia, typically over the internet.
16. **Blended Learning:** Combining traditional face-to-face instruction with online learning activities.

These pedagogies can be used individually or in combination, depending on educational goals, student needs, and subject matter. Each approach has its strengths and is suited to different learning objectives and classroom dynamics.




PRINCIPAL
Shreeyash Institute of Pharmaceutical
Education and Research, Aurangabad



Shreeyash Pratishthan's

SHREEYASH INSTITUTE OF PHARMACEUTICAL EDUCATION AND RESEARCH

Approved by AICTE, New Delhi, Government of Maharashtra, DTE Mumbai (DTE Code- 2572) and
Affiliated to Dr. Babasaheb Ambedkar Technological University, Lonere, & Maharashtra State Board
of Technical Education (MSBTE), (MSBTE Code- 1838) Mumbai

13/8/21

ACADEMIC PROGRAMME COMMITTEE

DUTIES:

- 1) Periodically reviewing the progress of the classes.
- 2) Discussing the problems concerning curriculum, syllabus and the conduct of classes.
- 3) Discussing with the course teachers on the nature and scope of assessment for the course and the same shall be announced to the students at the beginning of the respective semesters.
- 4) Communicating its recommendation to the head of the institution on academic matters.

MEETINGS:

The above class mentoring committee will meet before and after sessional examinations and whenever required.


CONSTITUTION:

The details of members are as follows:

Sr. No.	Name of Member	Designation
1	Dr. Ganesh G. Tapadiya	Chairman
2	Mr. Milind D. Kamble	Head of Department (Pharmaceutics)
3	Miss. Momin Aaliya Firdous	Academic In-charge
4	Mr. Mahendra Khandare	Examination Cell In-charge
5	Mrs. Arundhati Deokar	Member
6	Ms. Geetanjali Patil	Member
7	Mr. Mohammad Sayeed	Member
8	Mrs. Pallavi Bhosle	Member
9	Agnihotri Aakash (B. Pharm Final Year)	Student Representatives


Momin Aaliya Firdous
Academic In-Charge
Shreeyash Institute of Pharmaceutical
Education And Research
Aurangabad




Dr. Ganesh G. Tapadiya
PRINCIPAL
Shreeyash Institute of Pharmaceutical
Education and Research, Aurangabad.